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*MS 08-036
Environmental Toxicology*

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3 *Nanotechnology*

4 **Running head:** Effects of ingested nano-sized TiO₂ on *Porcellio scaber*

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13 Word Count: 6597

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Nanotechnology

EFFECTS OF INGESTED NANO-SIZED TITANIUM DIOXIDE ON TERRESTRIAL
ISOPODS *PORCELLIO SCABER*

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(Received 24 January 2008; Accepted 17 March 2008)

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66 **Abstract**–The effects of ingested nano-sized titanium dioxide (TiO₂, anatase, 15 nm) on
67 terrestrial isopods *Porcellio scaber* (Isopoda, Crustacea) after short-term (3 d) dietary exposure
68 were studied. Activities of antioxidant enzymes, such as catalase (CAT) and glutathione-S-
69 transferase (GST) in digestive glands were affected in a dose-independent manner, but higher-
70 level isopod endpoints, including weight change, feeding rate, food assimilation efficiency and
71 survival, were not affected up to the highest tested concentration of TiO₂ in the food (3000 µg/g
72 food). Exposure concentrations of 0.5, 2000, and 3000 µg of nonsonicated TiO₂/g food decreased
73 CAT and GST activities, but intermediate concentrations (1; 10; 100; 1000 µg/g food) did not
74 result in significant changes of enzyme activities. When the dispersion of TiO₂ was sonicated, no
75 effects on enzyme activities or higher level biomarkers were observed. The experimental set-up
76 with terrestrial isopods designed for dissolved chemicals is also suitable for testing the effects of
77 ingested nanoparticles, but the presentation of toxicity data needs to be adapted according to the
78 mode of action of nanoparticles and their specific characteristics.

79

80 **Keywords**–Nanoparticles, Feeding, Biomarkers, Terrestrial, Titanium dioxide

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87

INTRODUCTION

88

89 Bulk titanium dioxide (TiO₂) in micrometer-size dimensions has been in use for decades in
90 cosmetics, pharmaceutical, paint, and paper industries [1]. Titanium dioxide was labeled by
91 American conference of governmental industrial hygienists as a nuisance dust and it is thus
92 considered to be an inert dust not producing significant toxic effects under realistic exposures
93 [2]. Even though adverse effects of micrometer-sized TiO₂ have been demonstrated [3], TiO₂
94 traditionally is often considered to be nontoxic [4].

95 In the last decade, TiO₂ has been extensively produced in nano-sized form that has been used
96 increasingly in pollution treatment and remediation [4,5], disinfection, self-cleaning glass, solar
97 cells, electric devices, food additives, pharmaceuticals, and cosmetic products [6]. Nano-sized
98 TiO₂ is a well known photocatalyst. Namely, the TiO₂ crystalline forms are semiconductors,
99 meaning that they can be photo-excited to generate electron-hole pairs on its surface, which
100 results in their strong oxidizability [7]. This characteristic enhances the formation of reactive
101 oxygen species (ROS), which is among the main toxic mechanisms proposed for observed toxic
102 effects of photo-irradiated nano-sized TiO₂ [7].

103 Nano-sized TiO₂ without photo-activation has also been shown to cause adverse effects on a
104 variety of cell types, tissues and organisms. Examples include the cytotoxicity to rat lung
105 alveolar macrophages [8], human dermal fibroblast and human lung carcinoma cells [9],
106 apoptosis of Syrian hamster embryo fibroblasts [10], hepatic injury in mice [11], and pathologic
107 changes of gills in fish [12]. Similarly, it has been proposed, that the toxic mechanism of
108 nonirradiated nano-sized TiO₂ is ROS mediated [9, 10, 13]. Increased levels of hydrogen
109 peroxide, increased lipid peroxidation and decreased levels of reduced glutathione were observed

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110 in human bronchial epithelial cell line [13] and rat alveolar macrophages [8] exposed to
111 nonirradiated nano-sized TiO₂. The ability of nano-sized TiO₂ to induce ROS formation without
112 photo-activation has been related to its crystallinity and electronic configurations [14] and
113 indirect effect on the antioxidant system of the cell [15].

114 Because TiO₂ is classified as a dust, a majority of studies have been focused on its uptake by
115 and effects on the lungs [1]. However, since TiO₂ is used in food production, medicine, and
116 cosmetics, its oral ingestion is also an important exposure route [15]. Toxic effects of orally
117 ingested nano-sized TiO₂ have been demonstrated in mice [11] and rainbow trout [12].

118 Due to increasing introduction of nano-sized TiO₂ to the environment, it could potentially
119 provoke effects on a variety of organisms in different ecosystems. However, until recently the
120 majority of toxicity studies were focused on laboratory test mammals, such as rats and mice.
121 During the last two years, studies on the effects of nano-sized TiO₂ on aquatic organisms have
122 been performed [12, 16-18], but terrestrial toxicity studies are still lacking.

123 Currently, it is believed that a link between lower and higher level responses in test
124 organisms will provide the most relevant toxicity data for different organisms. Responses at
125 lower dose levels (biochemical biomarkers) can aid in the identification of the mechanisms
126 underlying the effects at higher levels of biological organization. Their disadvantage is however
127 often reflected in high variability of the response when compared to more integrated level
128 biomarkers [19].

129 The two biochemical biomarkers investigated in the present study catalase (CAT) and
130 glutathione-S-transferase (GST) are very highly conserved enzymes that have been identified in
131 most organisms, including vertebrates, invertebrates, plants, fungi, and bacteria [20]. The main
132 function of CAT is to catalyze the decomposition of hydrogen peroxide derived from the

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133 formation of other ROS such as superoxide or hydroxyl radical. Glutathione-S-transferase is a
134 member of a large family of multifunctional enzymes involved in the cellular detoxification of
135 many xenobiotics and physiological substances, including the endogenous products during lipid
136 peroxidation [20]. The activities of CAT and GST have been related mainly to antioxidant
137 function against ROS produced as a result of chemical stress, but they also have a central
138 metabolic function in the metabolism of ROS during normal cell functioning, where ROS appear
139 as side-products in a number of metabolic pathways [21, 22].

140 In cells stressed by exogenous chemicals, CAT and GST activities are reported to be
141 increased at lower concentrations of the chemical as a response to ROS production [20]. When
142 the chemical is present at higher concentrations, the decrease of enzyme activities has been
143 explained as a consequence of direct enzyme inhibition by the chemical, or as a result of cellular
144 dysfunction [23, 24].

145 In the present paper, the effects of ingested nano-sized TiO₂ on terrestrial isopod *Porcellio*
146 *scaber* (Isopoda, Crustacea) after a short-term (3 d) exposure are reported. The responses of
147 biomarkers at different levels of biological organization were measured to link the effects of
148 lower-level effects of ingested TiO₂ to responses at higher levels. Based on the reports that nano-
149 sized TiO₂ induces the production of ROS, we measured the activity of two biochemical
150 biomarkers CAT and GST [13, 14]. Weight change, feeding rate and food assimilation efficiency
151 of *P. scaber* were among conventional physiological parameters studied. Other studies provide
152 evidence that biochemical biomarkers are influenced by abiotic and biotic factors [25, 26].
153 Therefore the effects of laboratory rearing, moult stage, and gender on enzyme activities were
154 systematically evaluated prior to the main experiment with nano-sized TiO₂. We discuss the
155 effects of nano-sized TiO₂ on CAT and GST activities, and the suitability of the *P. scaber* test

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156 for assessing the effects of ingested nanoparticles.

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158

MATERIALS AND METHODS

159

Chemicals

161

162 The following chemicals were purchased from Sigma-Aldrich (Munich, Germany): dibasic
163 and monobasic potassium phosphate, hydrogen peroxide (30%), 1-chloro-2,4-dinitrobenzene, L-
164 glutathione (reduced form). Protein assay reagents A and B were purchased from Pierce
165 (Rockford, IL, USA). All chemicals were of the highest commercially available grade, typically
166 99% or higher.

167

Characterization of nano-sized TiO₂ particles

169

170 The nanoparticles of TiO₂ were supplied by Sigma-Aldrich in a form of a powder with 99.7%
171 purity. The following characteristics were provided by the supplier: anatase crystalline structure,
172 average particle size 15 nm, and surface area 190 to 290 m²/g.

173 Before applying TiO₂ to the leaves, different concentrations of TiO₂ were suspended in
174 bidistilled water with pH value of 5.7. The pH of the dispersions was independent on the
175 concentration of TiO₂ and was the same as in bidistilled water. The same bidistilled water was
176 used in control group and proved not to be toxic to isopods.

177 The sonicated and nonsonicated dispersions of TiO₂ were inspected with transmission
178 electron microscope analyses and dynamic light scattering technique (DLS). The dispersion

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179 prepared in bidistilled water (0.7 g/L) was sonicated for 30 min using 10 s pulses with 13872 J of
180 the total input of energy (Sonics vibra-cell, Ultrasonic processor VCX 750 Watt; Sonics &
181 Materials, Newtown, CT, USA). Both sonicated and nonsonicated dispersions were put on
182 carbon-coated grids, dried at room temperature and examined with a 200 keV field emission
183 transmission electron microscope (Philips CM 100, Koninklijke Philips Electronics, Eindhoven,
184 The Netherlands) and analyzed by transmission electron diffraction to determine the TiO₂ phase.

185 The same concentrations of sonicated and nonsonicated dispersions prepared in ultra-pure
186 water (Millipore, Billerica, MA, USA; ion free, pH = 5.7) (0.0066; 0.066; 0.222 g/L) were
187 inspected by DLS using a 3D-DLS-SLS Spectrometer (LS Instruments, Fribourg, Switzerland).
188 Other concentrations of TiO₂ prepared for toxicity experiments were also inspected by DLS, but
189 the measurements were not possible, because the signal was either too weak (at 0.00066 g/L) or
190 too strong (at 0.6667 and 2 g/L of TiO₂). The DLS measurements were performed on the TiO₂
191 dispersions without the addition of 0.1% tetrasodium pyrophosphate as previously used by
192 Warheit et al. [16], to mimic the composition of the dispersion used in toxicity tests.

193
194 *Effects of laboratory conditions on GST and CAT activities*

195
196 The effects of laboratory conditions on the CAT and GST activities of isopods collected from
197 the field were investigated. Animals were brought to the laboratory from three different field
198 locations in Slovenia: two sites near Domžale (vicinity of Ljubljana, Slovenia) (locations 1 and
199 2), and Radlek (Kozjansko, Slovenia) (location 3). All three locations have previously been used
200 as reference sites for isopod collection in toxicity studies, and were found uncontaminated [27,
201 28]. The activities of both enzymes were determined randomly after different periods (up to 55

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202 d) of culturing in the laboratory. The number of animals analyzed at different times is given in

203 **Table 1.** As a reference, a laboratory culture originating from a completely different location and

204 held in the laboratory for one year was also analyzed for enzyme activities.

205

206 *Exposure of P. scaber to TiO₂*

207

208 *Test organisms.* Terrestrial isopods (*Porcellio scaber*, Latreille 1804) were collected under

209 the litter layer in an uncontaminated location in the vicinity of Ljubljana (location 1). No

210 significant differences in enzyme activities of the animals analyzed within 1 to 3 d after the

211 collection from the field were found between the three locations mentioned above (**Fig. 1**),

212 therefore location 1 was selected as a source of animals for toxicity tests because of the close

213 proximity to the laboratory. In the laboratory, the animals were kept in a terrarium (20 × 35 × 20

214 cm) filled with a 2 to 5 cm layer of moistened sand and soil and a thick layer of partly

215 decomposed hazelnut tree leaves (*Corylus avellana*). The substratum in the terrarium was heated

216 to 80°C for several hours to destroy predators (spiders) before the introduction of the isopods.

217 The culture was kept at controlled room temperature (21±1°C), 16:8 h light:dark regime and

218 high humidity.

219 *Food preparation.* Hazelnut tree leaves were collected in an uncontaminated woodland and

220 dried at room temperature. Dry leaves were cut up into pieces of similar surface area, and

221 weighed. Pieces of approximately 100 mg were selected for the experiments. Prior to the

222 application of the TiO₂ dispersion, the leaves were kept in humid Petri dishes to facilitate the

223 absorption of applied TiO₂ dispersion. Afterwards, the leaves were dried for 24 h at room

224 temperature. The leaves were not directly exposed to light with the intensity of the illumination:

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225 16 h with 350 lux, and 8 h with 10 lux. The periods of maintenance of leaves in humid
226 environment and conditions for drying of the leaves were the same in all experiments.
227 The TiO₂ was suspended in bidistilled water using a vortex (20 s, 2000 rpm) and prepared freshly
228 each time prior to the experiment. No surfactants were used to disperse the TiO₂, since previous
229 studies have shown that dispersion using solely sonication is adequate [12, 16]. We applied 150
230 µl of the dispersion per 100 mg of leaf onto the lower leaf surfaces. Prior to the pipetting, the
231 dispersion was each time rotated on vortex for 5 s. Two types of TiO₂ dispersions were applied
232 onto the leaves; nonsonicated and sonicated. The sonicated dispersion was prepared using a
233 sonicator (30 min, 10 s pulses; Sonics vibra-cell, Ultrasonic processor VCX 750 Watt; Sonics &
234 Materials). Animals in the control group were fed with the leaves prepared in the same way, but
235 treated with the bidistilled water only.

236 *Experimental design.* The adults of *P. scaber* with body weights ranging from 30 to 80 mg,
237 of both sexes and all moult stages, were exposed to TiO₂ within 1 to 11 d after the collection in
238 the field. Each animal was placed individually in a Petri dish, to which individual pieces of dry
239 leaves dosed with TiO₂ were added. Humidity in the Petri dishes was maintained by regular
240 spraying with tap water on the internal side of the lids. All Petri dishes were placed in a large
241 plastic-covered glass container maintained at relative humidity close to 100%, and a 16:8 h
242 light:dark regime without the direct proximity of the lamp (illumination 16 h with 15 lux, and 8 h
243 with 5 lux).

244 After 3 d of exposure, lower and higher level endpoints were evaluated according to the test
245 protocol (**Table 2**). The animals and leaves were weighed after drying at room temperature for
246 24 h, and the faecal pellets were counted and weighted after drying in the exsiccator for 48 h.
247 The animals were dissected and the digestive glands (hepatopancreas) were isolated for

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248 measurements of CAT and GST activities. Animal mortality was also recorded.

249 Three separate experiments with a different range of tested concentrations were performed
250 (**Table 3**). Since currently no data exists on the environmental concentrations of nano-sized
251 TiO₂, the concentrations of TiO₂ used in the present study were selected based on preliminary
252 long-term studies, in which the effects on feeding and enzyme activities were observed at 1000
253 µg of TiO₂/g leaf (Drobne Damjana, University of Ljubljana, Biotechnical Faculty, Slovenia,
254 unpublished data, preliminary study). In experiment A, the isopods were exposed to high
255 concentrations (1000, 2000, 3000 µg/g leaf) of a nonsonicated dispersion of TiO₂. This
256 experiment was repeated three times, using ten animals per concentration each time. In
257 experiment B, the same high concentrations of sonicated dispersion of TiO₂ were tested and one
258 additional group was exposed to the highest tested concentration (3000 µg/g leaf) of
259 nonsonicated dispersion of TiO₂. This experiment was repeated twice, each time with ten
260 animals per concentration. In experiment C, the animals were exposed in three repeated
261 experiments to various very low concentrations (0.1, 0.5, 1, 10, 100 µg/g leaf) (C1, C2, C3) of a
262 nonsonicated dispersion of TiO₂. The number of animals and concentrations tested in each
263 repetition are described in Table 3.

264

265 *Determination of enzyme activities*

266

267 Animals of both genders and at all moult stages were used for enzyme analyses and a
268 separate enzyme sample was prepared from each animal. The whole digestive gland was
269 homogenized for 3 min in 0.8 ml of 50 mM phosphate buffer pH 7.0, using a Teflon®-glass
270 Elvehjem-Potter homogenizer (Cowie Technology, Middlesbrough, UK). The homogenate was

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271 centrifuged for 25 min at 15000 g and 4°C. The activities of both GST and CAT were measured
272 three times in each sample.

273 Glutathione-S-transferase activity was measured on microtiter plates (Bio-Tek[®] Instruments,
274 Winooski, VT, USA; PowerWave[™] XS) [29]. Final concentrations of both 1-chloro-2,4-
275 dinitrobenzene and reduced glutathione, prepared in 100 mM potassium phosphate buffer pH
276 6.5, were 1 mM. A detailed description of the preparation of 1-chloro-2,4-dinitrobenzene
277 solution is described in Jemec et al. [29]. We added 50 µl of the protein supernatant to start the
278 reaction, which was followed spectrophotometrically at 340 nm and 25°C for 3 min.

279 Glutathione-S-transferase activity was expressed in nmoles of conjugated reduced
280 glutathione/min/mg protein (extinction coefficient $\epsilon_{340} = 9600$ L/mol/cm).

281 Catalase activity was determined according to the method of Aebi [30]. We combined 100 µl
282 of protein supernatant with 700 µl of hydrogen peroxide solution (11.6 mM) in 50 mM potassium
283 phosphate buffer pH 7.0. The final concentration of hydrogen peroxide was 10.2 mM. The
284 reaction was followed spectrophotometrically for 3 min at 25°C and 240 nm on a Shimadzu
285 ultraviolet-2101PC spectrophotometer (Shimadzu, Kyoto, Japan). Catalase activity was
286 expressed in µmoles of degraded hydrogen peroxide/min/mg protein ($\epsilon_{240} = 43.6$ L/mol/cm).

287 Protein concentration was measured using a BCA[™] Protein Assay Kit, a modification of the
288 bicinchoninic acid protein assay (Pierce, Rockford, IL, USA).

289

290 *Data analysis*

291

292 Only animals between the two moults and females without brood chambers were included in
293 the analyses of higher level endpoints, because both the moult and the presence of broods might

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294 influence the feeding and animal mass change. The feeding rate and a defecation rate of isopods
295 were calculated as the mass of consumed leaf and mass of faecal pellets per animal wet weight
296 and per day, respectively. The food assimilation efficiency was calculated as the difference
297 between the feeding and defecation rates. The animal mass change was determined as the
298 difference in animal mass at the beginning and at the end of the experiment. The amount of the
299 daily consumed TiO₂ was calculated from the mass of consumed leaf and the corresponding
300 concentration of TiO₂ applied.

301 The significant differences between the control and exposed groups of animals were
302 determined by Kruskal-Wallis analysis and Mann-Whitney *U* test ($p < 0.05$) using Statgraphics
303 software (Statgraphics Plus for Windows 4.0, Statistical Graphics, Herndon, VA, USA).
304 Homogeneity of variance was tested using Levene's test.

305

306 RESULTS

307

308 *Characterization of nano-sized TiO₂ particles*

309

310 The transmission electron diffraction pattern revealed the TiO₂ to be in its anatase phase. The
311 TiO₂ nanoparticles were homogeneous in shape and size, in average 15 nm in diameter, and with
312 up to 1 to 5 aspect ratio between the diameter and length forming elongated spheroidal shapes
313 (**Fig.1**). They were strongly agglomerated. The comparison of sonicated and nonsonicated
314 samples did not reveal any effect of sonication on the intensity of agglomeration. The
315 nanoparticles might also agglomerate during evaporation of water in the process of transmission
316 electron microscope sample preparation, but on average, those clusters are more planar than

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317 those formed or retained in ultrasound agitation.

318 The DLS analyses on the other hand revealed the difference in the size of agglomerates after
319 sonication procedure. The median particle size of TiO₂ in sonicated aqueous dispersion was 350
320 to 500 nm, and 780 to 970 nm in nonsonicated dispersion.

321

322 *Effects of laboratory conditions on GST and CAT activities*

323

324 The activities of both enzymes gradually decreased during the culturing of isopods in the
325 laboratory. After three to four weeks of culturing in the laboratory, the span of enzyme activities
326 became narrow, remaining at a certain level and higher enzyme activities, which were observed
327 in certain specimens during the first 3 d, were no longer detected. After one year, the GST
328 activities of animals from all the tested locations were the same as after three to four weeks of
329 culturing in the laboratory. The same trend was observed for CAT activity at location 1, but at
330 locations 2 and 3 the activity of CAT was significantly higher in the reference laboratory culture
331 kept for one year in the laboratory than in those tested after 40 or 55 d (**Fig. 2**).

332 Enzyme activities of animals analyzed after 1 to 3 d, and 5 to 14 d were grouped to enable
333 comparison with the values obtained for control animals from toxicity experiments, which were
334 all analyzed within 4 to 14 d of the collection from the environment (Table 1).

335

336 *Variability of GST and CAT activities*

337

338 The CAT and GST activities in control animals from different toxicity experiments were
339 grouped in order to investigate the normal range of variability for these biochemical biomarkers

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340 and their dependence on gender, moult and presence of brood chamber. The animals included
341 were dissected within 4 to 14 d after the collection from the field.

342 Relatively high coefficients of variability for CAT and GST activities were observed (63.2
343 and 41.8%, respectively). The range of activities for control isopods were 0 to 55 $\mu\text{mol}/\text{min}/\text{mg}$
344 protein, and 83 to 624 $\text{nmol}/\text{min}/\text{mg}$ protein for CAT and GST, respectively (**Fig. 3**). No
345 statistically significant differences in the CAT and GST activities were found between males and
346 females, between animals in the intermoult and moult stage, and between females with and
347 without brood chamber.

348

349 *Effects of TiO₂ on P. scaber*

350

351 *Effects of TiO₂ on physiological endpoints.* After 3 d of exposure to TiO₂, no statistically
352 significant effects were observed on the feeding rate, defecation rate, food assimilation
353 efficiency, weight change and mortality of *P. scaber*. When higher level endpoints were
354 compared, no differences in the effects of TiO₂ between nonsonicated and sonicated TiO₂ were
355 observed.

356 Daily consumed levels of TiO₂ were calculated based on consumed food. No differences in
357 the quantity of consumed sonicated and nonsonicated TiO₂ were observed (**Fig. 4**).

358

359 *Effects of TiO₂ on enzyme activities.* Both enzyme activities were found to be decreased at
360 2000 and 3000 $\mu\text{g}/\text{g}$ leaf of nonsonicated TiO₂ (experiment A) (**Fig. 5 a,b**), but no changes were
361 observed in animals exposed to sonicated TiO₂ (experiment B) (**Fig. 5 c,d**). The decrease in
362 activity of both enzymes was also observed at 3000 $\mu\text{g}/\text{g}$ leaf of nonsonicated TiO₂, in the

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363 experiment B (Fig. 5 c,d). In experiment C, with low concentrations of TiO₂, the activities of
364 both enzymes were decreased at 0.5 µg/g leaf and again at 3000 µg/g leaf of nonsonicated TiO₂
365 (Fig. 5 e,f).

366 Based on the changes of CAT and GST activities presented above, it was concluded that the
367 response of both enzyme activities was independent of the dose of ingested nano-sized TiO₂.
368 This response is referred to here as binary. The relationship between the consumed levels of TiO₂
369 and the response of enzyme activities is shown in **Figure 6**.

370

371

DISCUSSION

372

373 The ingested nano-sized TiO₂ affected the activities of two antioxidant enzymes, CAT and
374 GST, in the digestive glands (hepatopancreas) of terrestrial isopods, *Porcellio scaber*, after 3 d of
375 exposure. The response was not dose-dependent. Only exposure concentrations 0.5, 2000, and
376 3000 µg/g of nonsonicated TiO₂ (corresponding to 0.066, 0.389, and 0.685 mg/kg body wt of
377 total consumed TiO₂, respectively), caused a decrease in the CAT and GST activities;
378 intermediate concentrations failed to provoke significant changes. Higher level responses like
379 feeding rate, defecation rate, food assimilation efficiency, weight change or mortality were not
380 affected up to 3000 µg/g of nonsonicated TiO₂ in the food. When the dispersion of TiO₂ was
381 sonicated prior to application, no effects on enzyme activities or higher level biomarkers were
382 observed.

383 We observed a large range of CAT and GST activities in the hepatopancreas of isopods
384 brought directly from the field. During the cultivation of isopods in the laboratory, these two
385 enzyme activities gradually decreased, and after approximately four weeks remained at a given

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386 level for a longer period. These changes could be explained as a part of a mechanism of
387 acclimation to laboratory conditions. Isopods in the natural environment are constantly exposed
388 to heterogeneous abiotic and biotic conditions, which result in the need for a pool of enzymatic
389 and nonenzymatic compounds, including CAT and GST, related to antioxidant defense. The
390 animals kept in the laboratory are exposed to a less variable environment may subsequently
391 result in a lower and narrower range of antioxidant enzyme activities. The effects of natural
392 factors on the activities of antioxidant enzymes are known [26]. In mussels (*Mytilus*
393 *galloprovincialis*) for instance, a high seasonal variation in antioxidant enzymes was reported
394 [25]. Oxidative changes have been related to changes in metabolic activity of the organisms, and
395 these changes depend on the intensity of feeding, temperature, and reproduction stage [21, 22].

396 We found no dependence of CAT and GST activities on the gender, moult stage or presence
397 of neonates in the brood chamber of *P. scaber*. Similarly, no links were found between the
398 gender and GST activities of *P. scaber* and *O. assellus* [31] and CAT activities and gender of the
399 marine amphipod *Gammarus locusta* [32]. In other organisms, some inconsistent results have
400 been reported. The CAT activities of mosquitofish (*Gambusi holbrooki*) are higher in males [33],
401 CAT activities in marine shrimp *Aristeus antennatus* are higher in females [34], and GST
402 activities in goodeid fish (*Girardinichthys viviparus*) are higher in males [35]. From these data, it
403 would appear that the effect of gender on CAT and GST activities is species-specific.

404 Hepatopancreatic CAT and GST activities in isopods fed with nonsonicated TiO₂ for 3 d
405 decreased in a dose-independent manner. Only 0.5, 2000, and 3000 µg/g of nano-sized TiO₂ in
406 the food-reduced CAT and GST activities; the intermediate concentrations (1; 10; 100; 1000
407 µg/g TiO₂) had no effect. It is evident that the observed changes of enzyme activities do not
408 depend on the concentration of TiO₂ nanoparticles, but are related to other properties of the

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409 nanoparticles. From the present study, it is impossible to deduce the explanation for observed
410 phenomena. However, at very high concentrations of TiO₂, the effect was similar to that of any
411 other dissolved chemical at high doses. The organism cannot cope with large amounts of
412 nanoparticles or dissolved chemicals, and the enzyme activities are affected as a result of the
413 impact on the general physiological state. Why low doses of TiO₂ also affected the two enzymes,
414 remains a challenge for further work.

415 Different relationships between the concentration of nano-sized TiO₂ and its effects have
416 been reported previously. For example, more pronounced effects at lower concentrations of
417 nano-sized TiO₂ (20 nm) have been reported for micronuclei sister chromatid exchanges in
418 Chinese hamster ovary cells [10]. An increase of algal growth was observed at 0.01, 0.1, and 10
419 mg/L of TiO₂ (140 nm), followed by a decrease at 100 mg/L [16]. No relationship between the
420 dose and immobility of daphnids was observed when the latter were exposed to TiO₂ (25 nm) in
421 a range from 1 to 3 mg/L [17]. On the other hand, a clear relationship between the dose and the
422 response was observed in different cytotoxicity assays with human cell line A549 cells at
423 concentrations of 0.3, 3, 30, 300, 1500, and 3000 µg/ml of nano-sized TiO₂ [9].

424 Because of the unique surface properties of nanoparticles, mechanisms of toxic action are
425 suggested which are distinct from the effect of soluble chemicals [1, 36]. In studies of
426 mechanism of action, Oberdörster et al. [37] have recommended a careful selection of tested
427 concentrations of nanoparticles. The importance of studying low concentrations of nanoparticles
428 to detect the possible hormetic response was emphasized. In the present study, the changes of
429 enzyme activities were detected at low concentrations, but the observed response was not
430 stimulatory as is typical for hormesis [38].

431 One of the primary problems in toxicity studies with nanoparticles is the preparation of

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432 exposure medium, since the particles tend to form aggregates in water. Most commonly, the
433 dispersion of nano-sized TiO₂ is sonicated to diminish the aggregation of particles [9]. The
434 control of the size of aggregates delivered to terrestrial isopods is impossible, because the
435 preparation of the food involves drying of the stock dispersions on the leaves, which results in
436 changes in aggregate formation. Furthermore, the TiO₂ ingested by isopods can be further
437 (dis)aggregated in the digestive system where a pH is 6 to 6.5 and high concentration of
438 surfactants is present [39]. Until no standard protocol is established for terrestrial toxicity studies
439 of nanoparticles, we recommend the sonication of the stock dispersions prior to application onto
440 food, because it enables the application of more uniform aggregates than without sonication.

441 The binary response of CAT and GST activities obtained in the present study differs from the
442 typical dose–response described for soluble chemicals and consequently, the determination of
443 toxicological data, such as the no- or lowest-observed-effect concentration (level) (NOEC,
444 NOEL, LOEC, LOEL) or effect concentrations (EC_x, EL_x, etc.) is not possible. No similar
445 reports for nano-sized TiO₂ are found in the literature and it will be interesting to see whether the
446 binary response is also a characteristic of other nanoparticles and endpoints when a wide range
447 of concentrations is tested.

448 Previous ecotoxicity studies with aquatic test organisms showed that nano-sized TiO₂
449 (aggregates in water 140 nm) exhibit low concern for the immobility of rainbow trout
450 *Oncorhynchus mykiss* and water fleas *Daphnia magna* (no effect up to 100 mg/L) and medium
451 concern for the growth of algae *Pseudokirchneriella subcapitata* (72 h median effective
452 concentration EC₅₀ = 87 mg/L) [16]. Similarly, no effect on daphnid immobility was observed
453 up to 500 mg/L of sonicated nano-sized TiO₂ (aggregates in water 100-500 nm) [18]. The results
454 of the present study imply, that nano-sized TiO₂ after 3-d of exposure is less toxic for the feeding

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455 rate of isopods than some dissolved metals, for instance ZnCl_2 . Namely, our unpublished data
456 show, that the feeding rate of isopods is affected already after 3-d of feeding with 2000 $\mu\text{g/g}$ of
457 Zn^{2+} , while no effect of 3000 $\mu\text{g/g}$ of TiO_2 was observed in the present study after the same
458 exposure period. After 14 d of exposure, the feeding of isopods was affected at 125 $\mu\text{g/g}$ of Cd^{2+} ,
459 1200 $\mu\text{g/g}$ of Cu^{2+} and 1800 $\mu\text{g/g}$ of Zn^{2+} [28]. Further studies on the effects of the same nano-
460 sized TiO_2 on isopods after 14-d of exposure are under preparation and will clarify the relative
461 toxicity of nano-sized TiO_2 for isopods.

462 Data on currently detected environmental concentrations of nano-sized or bulk TiO_2 in the
463 terrestrial or aquatic environment are not available, since TiO_2 , not included in the priority list of
464 toxic pollutants in European Union ([40];
465 http://www.mepa.org.mt/index.htm?eu_int_affairs/eu_legislation/mainpage.htm&1), is not
466 systematically monitored. According to the predicted future use of various nanoparticles,
467 including TiO_2 , they will undoubtedly enter the environment either via domestic or industrial
468 wastewaters or direct use for the removal of pollutants from contaminated water and soil, use in
469 water treatment filters and control of algal growth in water systems. Warheit et al. [16] recently
470 reported a minimum base set of toxicity tests used for nanoparticle risk management, including
471 pulmonary toxicity studies, acute dermal toxicity and sensitization studies, acute oral and ocular
472 toxicity studies, genotoxicity studies, and aquatic toxicity studies. Due to potential introduction
473 of nanoparticles in the soil, the inclusion of a set of terrestrial toxicity tests in risk
474 characterization of nanoparticles is necessary.

475 The advantage of toxicity tests using the terrestrial isopods lies in the possibility to determine
476 the consumption levels of chemicals and subsequently the effective dose. We recommend the use
477 of the same experimental design both for dissolved chemicals and for nanoparticles, but the

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478 presentation of toxicity data needs to be adapted to reflect the nanoparticles' mode of action and
479 specific characteristics. Since virtually no agreement currently exists as to how to present the
480 toxicity data for nanoparticles, such data will probably have to be evaluated retrospectively and it
481 is therefore very important to provide as much supplementary data as possible. The data should
482 include, for example, a detailed description of organisms (life span, different physiological
483 states, gender, health status), a detailed description of the preparation of dispersions of
484 nanoparticles, characterisation of particles, exposure concentrations or ingested dose per day,
485 total ingested dose, duration of exposure, a range of tested concentrations, and as much toxicity
486 data as possible. Such information can be derived from toxicity testing with terrestrial isopods.

487

488

CONCLUSION

489

490 The results of the present study show that hepatopancreatic activities of CAT and GST
491 gradually decrease during the cultivation of isopods (*Porcellio scaber*) in the laboratory, and that
492 the activities of these two enzymes are not dependent on gender, presence of moult or
493 marsupium. After short-term (3 d) dietary exposure of terrestrial isopods to 15 nm TiO₂ (anatase)
494 the activities of CAT and GST were affected in a dose-independent manner, higher level
495 responses of isopods were not changed, and the sonication procedure of TiO₂ dispersion altered
496 its toxic potential for enzyme activities. The presented experimental set-up with terrestrial
497 isopods was found to be suitable for testing the effects of ingested nanoparticles and is
498 recommended in future risk characterization of nanoparticles.

499

500 *Acknowledgement* –The present study received financial support from Slovenian Research

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501 Agency and the Slovenian Science Foundation (World Federation of Scientists National

502 Scholarship). We thank Ksenija Kogej for DLS analysis.

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639 **FIGURE LEGENDS:**

640 **Fig. 1.** Transmission-electron micrographs of nanosized titanium dioxide (TiO₂) in bidistilled
641 water. Elongated spheroidal shapes of TiO₂ nanoparticles with an average diameter 15 nm and
642 with up to 1 to 5 aspect ratio between the diameter and length (left), and the typical agglomerates
643 of nanoparticles (right) are shown.

644
645 **Fig. 2.** The effect of laboratory conditions on catalase (CAT) and glutathione-*S*-transferase
646 (GST) activities in isopods (*Porcellio scaber*) collected from three different field locations:
647 location 1 (Ljubljana, Slovenia) (**a,d**) location 2 (**b,e**) (Ljubljana, Slovenia) and location 3
648 (Radlek, Slovenia) (**c,f**). The laboratory culture (lab. culture) originating from another location
649 was analyzed as a reference after one year. The data for all animals and the means of a certain
650 time group (black thick line) are shown. Statistically significant differences compared to the first
651 time group (1-3 d) (*), and statistically significant differences between the laboratory culture (1
652 year) and the preceding time group (#) are shown ($p < 0.05$).

653
654 **Fig. 3.** The variability of catalase (CAT) (**a**) and glutathione-*S*-transferase (GST) (**b**) activities of
655 control animals and their dependence on the physiological state of isopods (*Porcellio scaber*).
656 Symbols on the box plot represent: maximum and minimum value (whiskers: ⊥), mean value
657 (■), females with brood chamber (BC), and females without brood chamber (no BC) ($p < 0.05$).

658
659 **Fig. 4.** Daily consumed levels of nano-sized titanium dioxide (TiO₂) in all three experiments (A,
660 B, and C). Symbols on the box plot represent maximum and minimum value (whiskers: ⊥), and
661 mean value (■). (*n* = nonsonicated TiO₂, *s* = sonicated TiO₂, // on y-axis = a brake).

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662

663 **Fig. 5.** The effects of high concentrations of nonsonicated nano-sized titanium dioxide (TiO₂)
664 **(a,b)** (experiment A), sonicated TiO₂ **(c,d)** (experiment B), and low concentrations of
665 nonsonicated TiO₂ **(e,f)** (experiment C) on catalase (CAT) and glutathione-S-transferase (GST)
666 activities in digestive glands of *Porcellio scaber* after 3 d of exposure. Symbols on the box plot
667 represent: maximum and minimum value (whiskers: ⊥), mean value (■), significant changes
668 compared to control (*) ($p < 0.05$), (n = nonsonicated TiO₂) (graphs c,d). The dashed line
669 represents the mean value of the control.

670

671 **Fig. 6.** The relationship between the changes of catalase (CAT) **(a)** (and glutathione-S-
672 transferase (GST) **(b)** activities and the amount of a daily consumed nano-sized titanium dioxide
673 (TiO₂). The data presented are from the experiment C (mean ± standard errors of the mean). The
674 enzyme activities are expressed as a percentage compared to a mean control value. The dashed
675 horizontal line represents the control (100%).

676

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Table 1: The number of animals analyzed after a certain period of cultivation in the laboratory (Fig.1)

Days in the laboratory	LOCATIONS		
	1	2	3
	Number of animals analyzed		
1	9	9	13
3	7	7	
5	7	6	
8		7	
10	7		
14	8	7	9
17			7
21		7	
28			7
30	6	7	
40		6	
45	7		
55			5

Table 2: Overview of the testing protocol

Test organism description		Endpoints evaluated
Invertebrate Isopoda, Crustacea Terrestrial isopod <i>Porcellio scaber</i>		<i>Lower level end-points:</i> Digestive glands: - Glutathione-S-transferase activity
Type of exposure	Chemical	- Catalase activity
3 d dietary exposure	nano-sized titanium dioxide (anatase, 15 nm) -Sonicated -Nonsonicated	<i>Higher level end-points:</i> - Feeding rate - Defecation rate - Food assimilation efficiency - Animal mass change - Mortality

Table 3: Experimental set-up

Suspension of TiO ₂	Final concentrations of titanium dioxide (TiO ₂) on leaves (µg/g leaf)	Experiments					
		Total No. of exposed animals					
		A	B	C			
Nonsonicated				C 1	C 2	C 3	ΣC
	0	30	20	6	8	10	24
	0.1				10	10	20
	0.5				10	10	20
	1			6	8		14
	10			6			6
	100			6	7	10	23
	1000	30		6			6
	2000	30					
	3000	30	20		7		7
Sonicated	1000		20				
	2000		20				
	3000		20				

ΣC- total amount of animals in experiment C.

Fig. 1.

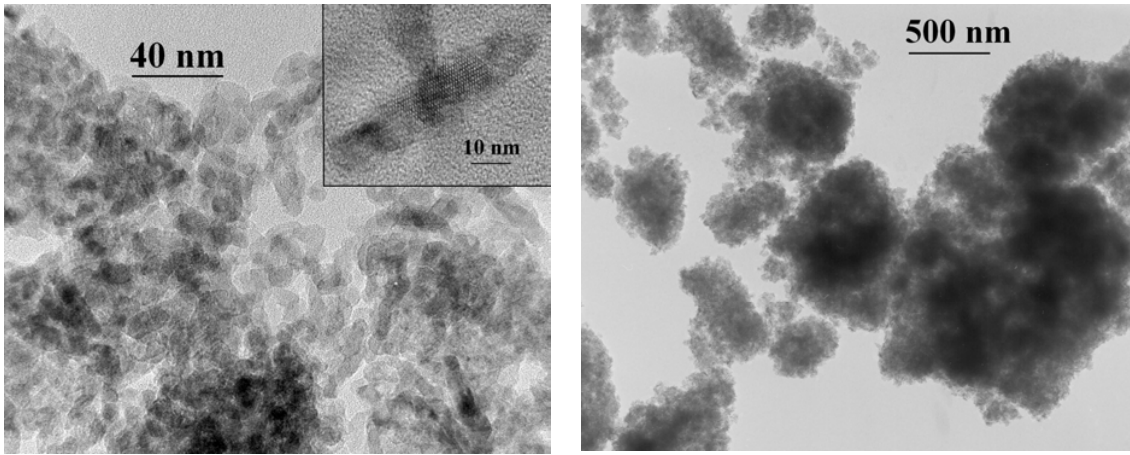
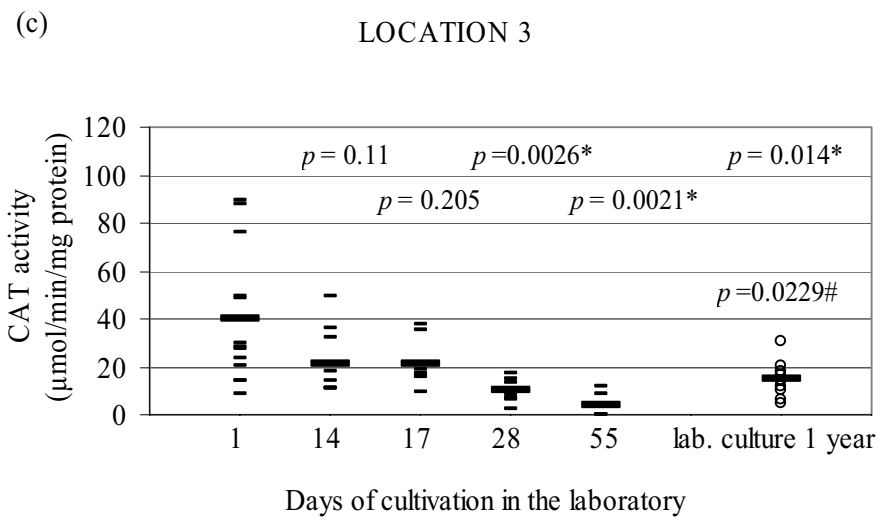
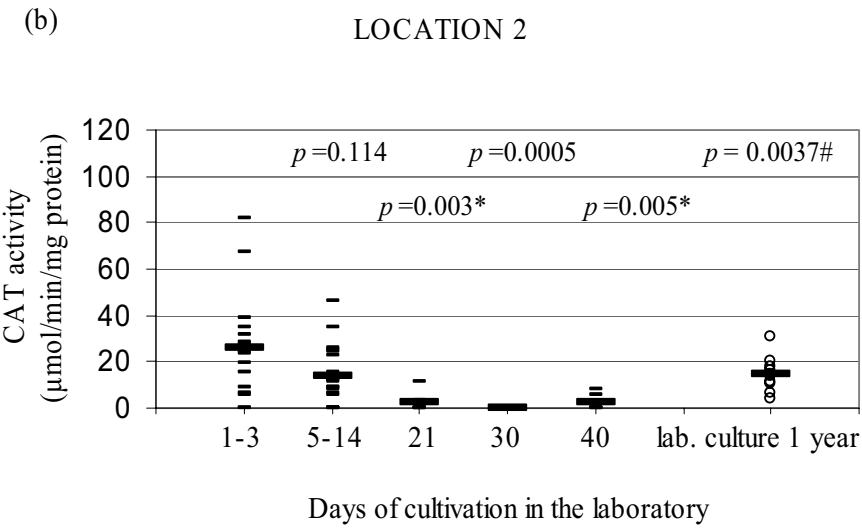
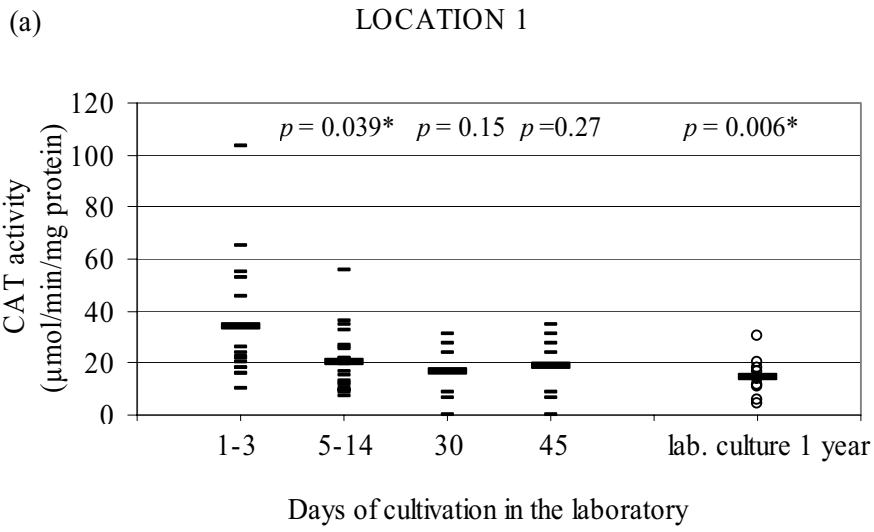
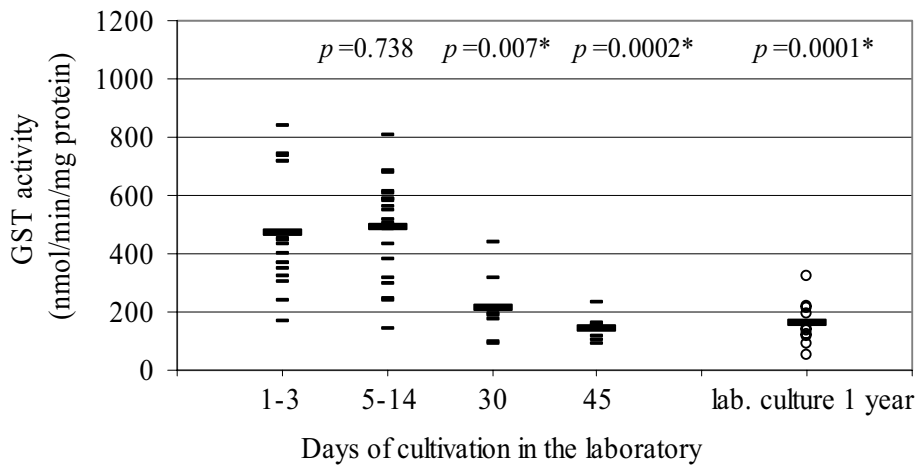


Fig. 2.



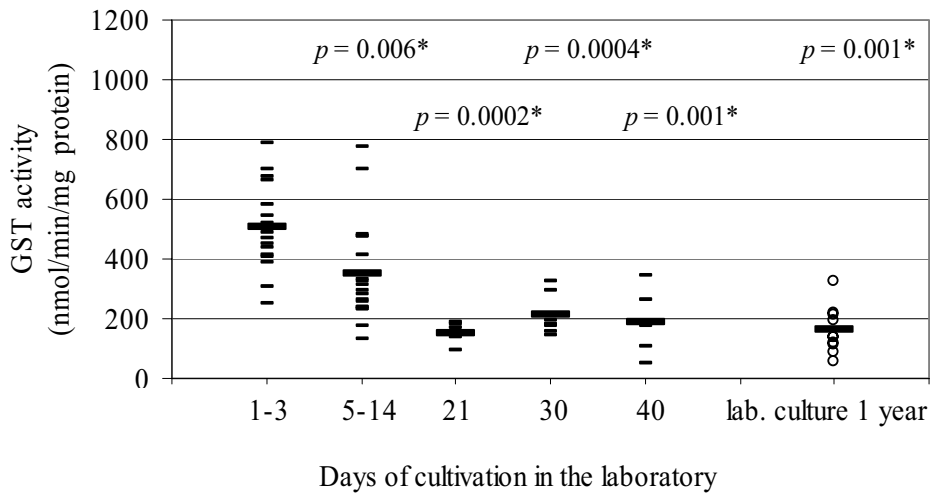
(d)

LOCATION 1



(e)

LOCATION 2



(f)

LOCATION 3

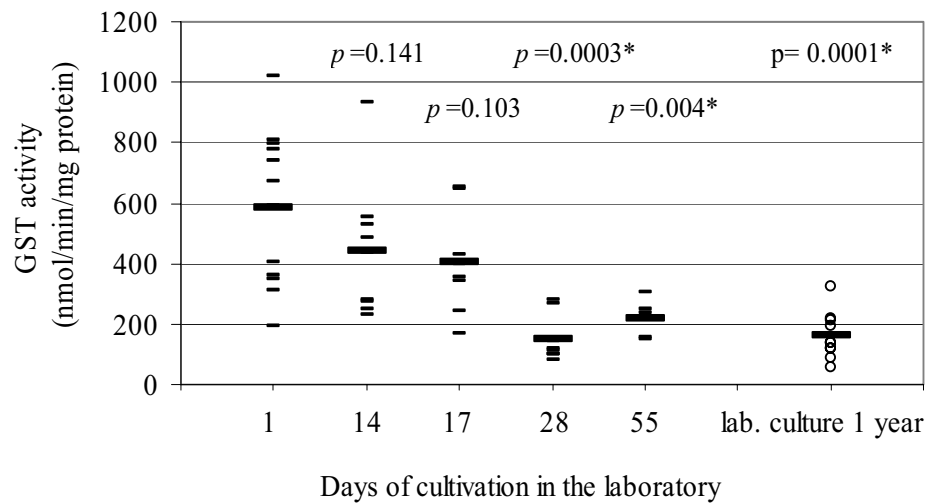
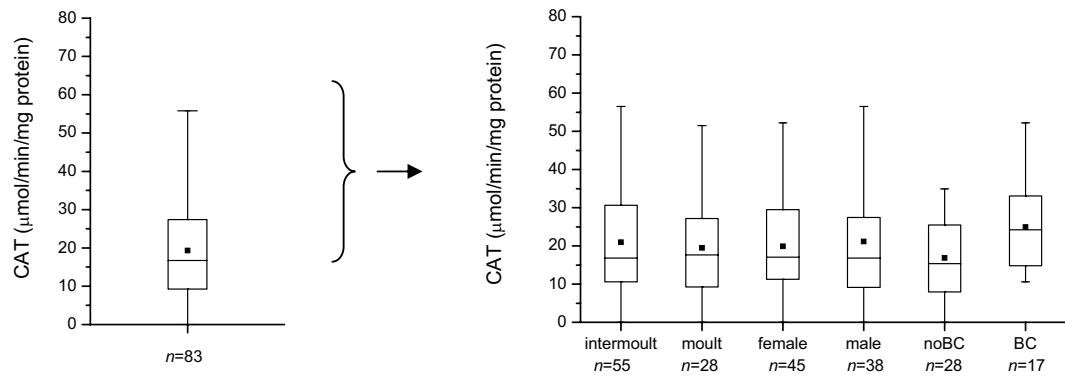


Fig. 3.

(a)



(b)

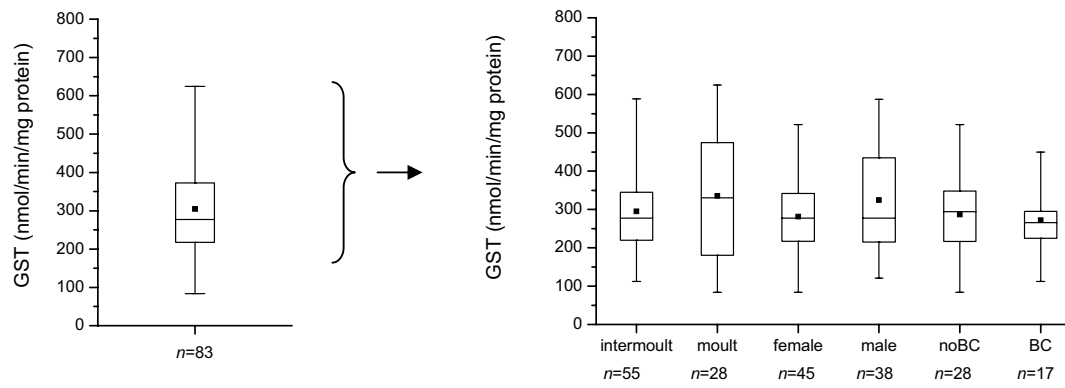


Fig. 4.

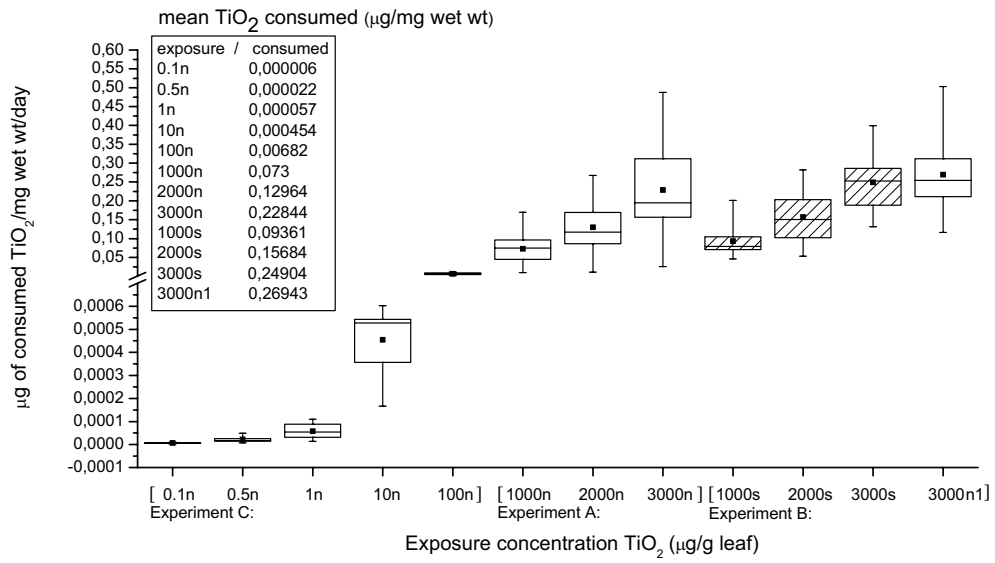


Fig. 5.

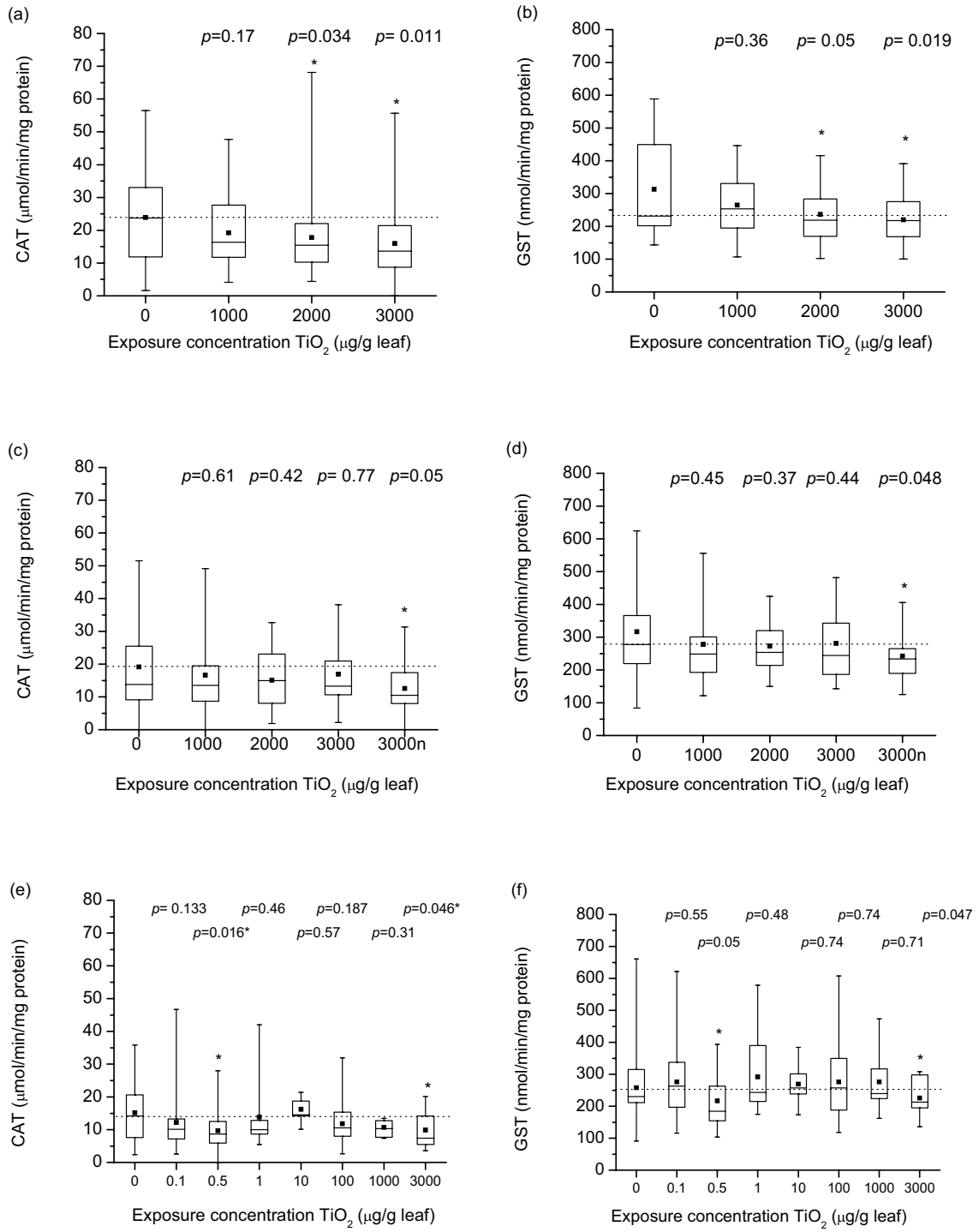


Fig. 6.

