

1 **In vivo Screening to Determine Hazards of Nanoparticles: Nanosized TiO₂**

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21

22 **Abstract**

23

24 A single-species laboratory test with terrestrial invertebrates was used to identify the
25 hazard of nanosized TiO₂. Feeding parameters, weight change, mortality, and the
26 activities of catalase and glutathione- S-transferase were evaluated after three or 14 days
27 of dietary exposure. The effects of nano-TiO₂ were dependent on exposure
28 concentration and duration, total consumed quantity, size and pre-treatment of particles.
29 The intensity of a response was ruled by duration of exposure and not by consumed
30 quantity of nano-TiO₂ or exposure concentration as expected. The response to nano-
31 TiO₂ is described as threshold-like. The exposure concentrations 10-1000 µg TiO₂/g
32 dry food (1.35-1025 µg of total consumed quantity of TiO₂/g animal wet wt) were
33 identified as safe for tested species after tested exposure period. We conclude that the
34 response to nanoparticles is different from that of soluble chemicals therefore these two
35 types of data should be interpreted and processed differently.

36

37 **Keywords:** Nanoparticles, Dietary exposure, Terrestrial invertebrates, Biomarkers,
38 Biological activity

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40 *Capsule: The response of a biological system to nanoparticles is unique and depends on*
41 *their physico-chemical characteristics, dose and duration of exposure.*

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46 1 INTRODUCTION

47 The ecotoxicity data on the effects of nanoparticles are in much need for the
48 appropriate environmental risk assessment. Different documents already exists which
49 deal with emerging and newly identified health risks (TGD Document, 2003; NANO
50 Risk Framework, 2007; SCENIHR, 2007). Development of a hazard profile is the
51 critical step in characterizing the potential safety of nanoparticles, and the associated
52 health and environmental hazards. A base set of hazard data has been suggested as a
53 reference for characterization and prioritization of nanoparticles (Warheit et al., 2007a).

54 To characterize nanoparticles and its potential hazards sufficiently, empirical
55 data are necessary. Since the early days of the REACH proposals (REACH, 2006), it
56 has been agreed by all partners that the number of animals used to gain toxicity
57 information on chemicals should be kept to an absolute minimum. There is evidence
58 that in vitro and in silico methods for acute chemical toxicity are able to provide
59 sufficient data to permit classification and labelling. However, for those substances with
60 no available toxicity data a read-across and quantitative structure-activity relationship
61 techniques (QSAR) are not possible, therefore in vivo testing is required to rapidly
62 identify hazardous substances. Tests with invertebrates are suitable for such purposes
63 since they are not subjected to the same legal restrictions as vertebrates.

64 We present a laboratory single-species toxicity test with the terrestrial arthropod
65 (*Porcellio scaber*, Isopoda, Crustacea) for the purposes of hazard identification of
66 nanosized TiO₂. The experimental design presented in this work provides data on
67 biological responses from several levels of biological organisation; e.g. lower level
68 (enzyme activities) and higher level (feeding, growth and mortality). The two enzymes
69 investigated in the present study were catalase (CAT) and glutathione- S-transferase

70 (GST), both of which are involved in antioxidant defence against reactive oxygen
71 species. The main function of CAT is to catalyze the decomposition of hydrogen
72 peroxide, while GST is a member of a large family of multifunctional enzymes involved
73 in the cellular detoxification of many xenobiotics and physiological substances,
74 including the endogenous products during lipid peroxidation. Our previous work has
75 shown that the advantage of this test is that it provides a variety of toxicity data based
76 on exposure concentration (such as lowest- and no-observed exposure concentration)
77 and also exposure dose (lowest- and no- observed exposure dose). The isopod toxicity
78 test system has been used successfully in metal and pesticide toxicity studies (Drobne,
79 1997; Stanek et al., 2006).

80 To validate our test system for testing of nanoparticles we selected
81 nanoparticulate matter for which some toxicity information already exists (Hund-Rinke
82 and Simon, 2006; Federici et al., 2007; Lovern and Klaper, 2006; Warheit et al., 2007a).
83 The nanosized TiO₂ has a number of industrial applications such as a food colouring,
84 additive in pharmaceuticals and cosmetics and, due to its photo-physical properties, it is
85 also used in a wide range of other consumer products (Masciangoli et al., 2003). It had
86 been considered biologically inert prior to studies with ultra-fine particles which
87 showed that ultra-fine TiO₂ particles (20 nm in diameter) provoked an inflammatory
88 response in laboratory test organisms (Oberdörster et al., 1994).

89 The aim of the present work was to investigate the hazard of nanosized TiO₂.
90 We investigated: (a) exposure duration-effect relationship; (b) exposure concentration
91 (dose)-effect relationship; (c) effect-particle size relationship; and (d) the effect of
92 nanoparticle pre-treatment. We compare our toxicity data on TiO₂ with literature reports

93 and discuss suitability of terrestrial isopods for hazard identification of engineered
94 nanoparticles.

95

96 **2 MATERIALS AND METHODS**

97

98 *2.1 Characterization of TiO₂ nanoparticles*

99

100 Two sizes of commercially available TiO₂ nanoparticles (Sigma-Aldrich) were
101 investigated: <25 nm in diameter (referred to here as ‘smaller’) and <75 nm in diameter
102 (‘larger’). The characteristics provided by the supplier are described in Table 1.
103 Additional characterisation of the test material was performed on the nanoparticles as
104 delivered (either powder or liquid medium) and also dispersed in bidistilled water (pH
105 value 5.7), which was used to prepare the food for isopod toxicity testing.

106 The commercial material was investigated by BET analysis (Brunauer- Emmett
107 -Teller surface area analysis; Tristar 3000, Micrometrics) (Braunauer et al., 1938) to
108 obtain information concerning the surface area of the solid material. Here, samples were
109 dried and degassed with nitrogen prior to analysis.

110 Sonicated and non-sonicated dispersions of TiO₂ in bidistilled water were
111 inspected by transmission electron microscopy (TEM) and a dynamic light scattering
112 technique (DLS). The dispersions prepared in bidistilled water (0.0066; 0.066; 0.667
113 g/L) were sonicated on ice for 30 min using 10 s pulses with 13872 J of the total input
114 of energy (Sonics vibra-cell, Ultrasonic processor VCX 750 Watt; Sonics & Materials,
115 Newtown, CT, USA). Both sonicated and non-sonicated dispersions were put on

116 carbon-coated grids, dried at room temperature and examined by TEM (Philips CM
117 100).

118 The same concentrations of sonicated and non-sonicated dispersions prepared in
119 ultra-pure water filtered through a 0.2 µm sieve (Millipore, Billerica, MA, USA; ion
120 free, pH = 5.7) were inspected by DLS using a 3D-DLS-SLS Spectrometer (LS
121 Instruments, Firbourg, Switzerland).

122

123 2.2 *Exposure of isopods P. scaber to TiO₂*

124

125 2.2.1 *Test organisms*

126

127 Terrestrial isopods (*Porcellio scaber*, Latreille 1804) were collected under the
128 litter layer in an uncontaminated location in the vicinity of Ljubljana. In the laboratory,
129 the animals were kept in a terrarium (20 × 35 × 20 cm) filled with a 2 to 5 cm layer of
130 moistened sand and soil and a thick layer of partly decomposed hazelnut tree leaves
131 (*Corylus avellana*). The substratum in the terrarium was heated to 80°C for several
132 hours to destroy predators (spiders) before the introduction of the isopods. The culture
133 was kept at controlled room temperature (21±1°C), 16:8 h light/dark regime and high
134 humidity. The adults of *P. scaber* of both sexes and with body weights ranging from 30
135 to 80 mg, and all moult stages, were exposed to TiO₂ within 1 to 14 d after collection in
136 the field. It has been previously shown, that these confounding factors do not influence
137 the possible toxic outcomes of pollutants on isopods (Jemec et al., 2008).

138

139 2.2.2 *Experimental design*

140

141 Each animal was placed individually in a Petri dish, to which individual pieces
142 of TiO₂-treated dry leaves were added. Humidity in the Petri dishes was maintained by
143 regular spraying with tap water on the internal side of the lids. All Petri dishes were
144 placed in a large plastic-covered glass container maintained at approximately 100%
145 relative humidity and a 16:8 h light/dark regime without the direct proximity of the
146 lamp (illumination 16 h with 203 nmol m⁻² s⁻¹ (15 lux), and 8 h with 67 nmol m⁻² s⁻¹ (5
147 lux).

148 After 3 d and 14 d of exposure, lower and higher level end-points were
149 evaluated according to the test protocol (Table 2). Animal mortality was recorded, the
150 surviving animals were weighed at the end of the experiments, and the leaves were
151 dried at room temperature for 24 h prior to weighing. Faecal pellets were counted and
152 weighed after drying in the exsiccator for 48 h. The animals were dissected and the
153 digestive glands (hepatopancreas) were isolated for measurements of catalase (CAT)
154 and glutathione S-transferase (GST) activities.

155 Altogether, four experiments were performed (Table 3). In two of them (A, B)
156 we assessed the effect of duration of exposure (3d and 14d). In other two (B, C) the
157 main focus was placed on the effect of particle size (<25 nm and <75 nm), and in one
158 simple experiment (D) we tested the possible influence of the pre-treatment of
159 nanoparticles. Based on previous results only one concentration (1000 µg of TiO₂/g dry
160 food) of small size nanoparticles was selected for this purpose. Exposure concentrations
161 presented in Table 3 are nominal concentrations, no actual concentrations on leaves
162 were measured.

163 The number of animals tested in each experiment depended on the abundance of
164 population collected prior to exposure (Table 3). Namely, our previous work has shown,
165 that the animals investigated for enzyme activities have to be collected at the same field
166 location and exposed immediately after the collection (Jemec et al., 2008).

167 Since currently no data exists on the environmental concentrations of nanosized
168 TiO₂, the concentrations of TiO₂ used in this study were selected based on preliminary
169 short-term studies, where the effects on enzyme activities were observed up to 3000 µg
170 of TiO₂/g dry food (Jemec et al., 2008).

171

172 2.2.3 Food preparation

173

174 Food was prepared as previously described (Jemec et al., 2008). Hazelnut tree
175 leaves were collected in uncontaminated woodland, dried at room temperature and the
176 dry leaves were cut up into pieces of similar surface area, and weighed. Pieces of
177 approximately 100 mg were selected for the experiments. Before the application of TiO₂
178 to the leaves, different concentrations of TiO₂ (0.0066; 0.066; 0.667 g/L) were
179 suspended in bidistilled distilled water with pH value of 5.7. The pH of the dispersions
180 was independent on the concentration of TiO₂ and was the same as in bidistilled water.
181 The TiO₂ was suspended using a vortex (20 s, 2000 rpm) and prepared freshly for each
182 experiment. Surfactants were not used to disperse the TiO₂, since previous studies have
183 shown that dispersion using solely sonication is adequate (Federici et al., 2007; Warheit
184 et al., 2007b). 150 µl of the dispersion per 100 mg of leaf were applied onto the lower
185 leaf surfaces and dispersed using a paintbrush. Dispersions of TiO₂ with concentrations
186 0.0066, 0.066 and 0.667 g/L resulted in final concentrations of 10, 100 and 1000 µg of

187 TiO₂/g dry food. Prior to sampling of the suspension, the dispersion was each time
188 rotated on a vortex for 5 s. Non-sonicated and sonicated dispersions of TiO₂ were
189 applied to the leaves. The sonicated dispersion was prepared using a sonicator (30 min,
190 10 s pulses; Sonics vibra-cell, Ultrasonic processor VCX 750 Watt; Sonics & Materials,
191 Newtown, CT, USA). Animals in the control group were fed with the leaves prepared in
192 the same way, but treated with the distilled water only.

193

194 2.2.4 Determination of enzyme activities

195

196 Animals of both genders and at all moult stages were used for enzyme analyses
197 and a separate enzyme sample was prepared from each animal. The whole digestive
198 gland was homogenized for 3 min in 0.8 ml of 50 mM phosphate buffer pH 7.0, using a
199 teflon–glass Elvehjem-Potter homogenizer. The homogenate was centrifuged for 25 min
200 at 15000 g and 4°C.

201 GST activity was measured on microtiter plates (Bio-Tek[®] Instruments,
202 Winooski, VT, USA; PowerWave[™] XS) (Habig et al., 1974; Jemec et al., 2007). Final
203 concentrations of both 1-chloro-2,4-dinitrobenzene and reduced glutathione, prepared in
204 100 mM potassium phosphate buffer pH 6.5, were 1 mM. A detailed description of the
205 preparation of 1-chloro-2,4-dinitrobenzene solution is has been described previously
206 (Jemec et al., 2007). 50 µl of the protein supernatant was added to start the reaction
207 which was followed spectrophotometrically at 340 nm and 25°C for 3 min. GST activity
208 was expressed in nmoles of conjugated reduced glutathione min⁻¹ mg protein⁻¹
209 (extinction coefficient, $\epsilon_{340} = 9600 \text{ L mol}^{-1} \text{ cm}^{-1}$).

210 Catalase activity was determined according to a published method (Aebi 1984).
211 100 μl of protein supernatant was combined with 700 μl of hydrogen peroxide solution
212 (11.6 mM) in 50 mM potassium phosphate buffer pH 7.0. The final concentration of
213 hydrogen peroxide was 10.2 mM. The reaction was followed spectrophotometrically for
214 3 min at 25°C and 240 nm in a Shimadzu ultraviolet-2101PC spectrophotometer
215 (Shimadzu, Kyoto, Japan). Catalase activity was expressed in μmoles of degraded
216 hydrogen peroxide $\text{min}^{-1} \text{mg protein}^{-1}$ ($\epsilon_{240} = 43.6 \text{ L mol}^{-1} \text{ cm}^{-1}$). The activities of both
217 GST and CAT were measured three times in each sample.

218 Protein concentration was measured using a BCA™ Protein Assay Kit, a
219 modification of the bicinchoninic acid protein assay (Pierce, Rockford, IL, USA).
220

221 2.3 Data analysis

222

223 At the end of experiment faecal pellets were removed completely from the leaves
224 using a brush, they were counted and weighted. Also, the leaves were weighted. The
225 feeding rate and defecation rate of isopods were calculated as the mass of consumed leaf
226 and mass of faecal pellets per animal wet weight per day, respectively. The food
227 assimilation efficiency was calculated as the difference between the mass of consumed
228 leaf and mass of faecal pellets divided by the mass of consumed leaf. The animal mass
229 change was determined as the difference in animal mass at the beginning and at the end
230 of the experiment. The amount of the daily consumed TiO_2 was calculated from the
231 mass of consumed leaf and the corresponding applied concentration of TiO_2 .

232 Homogeneity of variance was tested with Levene's test. The differences
233 between the control and exposed groups of animals were determined by Kruskal-Wallis

234 analysis and the Games-Howell post hoc test using SPSS for Windows 8.0 (SPSS Inc.,
235 USA). The comparison of data was done within a single experiment, no cross statistical
236 comparisons between the experiments (A-D) were performed.

237

238 **3 RESULTS**

239

240 **3.1 Characterization of nanosized TiO₂ particles**

241

242 The characteristics of nanosized TiO₂ are provided in Table 1. The BET analysis
243 revealed that both sizes of nanosized TiO₂ formulations were in accord with the data
244 provided by supplier (smaller < 25 nm, larger < 75 nm). BET revealed a specific surface
245 area of 145 m²/g for the small sized TiO₂ nanoparticles and 40 m²/g for larger sized
246 nanoparticles.

247 The TEM analysis showed that looser aggregates of nano-TiO₂ were formed
248 when the dispersion was sonicated in comparison to non-sonicated small sized TiO₂.
249 Looser aggregates were also formed in the case of larger nanosized TiO₂ in comparison
250 to smaller one (Fig.1).

251 Similarly, the size of aggregates as determined by DLS (0.0066 and 0.066 g/L of
252 TiO₂) was lower in the case of sonicated smaller TiO₂ and larger TiO₂ form in
253 comparison to non-sonicated small sized TiO₂. Concentrations of 0.667 g/L of TiO₂
254 were not examined by DLS, because at such high concentrations the signal was beyond
255 the scale of the detector.

256

257 **3.2 The effects of nanosized TiO₂ on *P. scaber***

258

259 The results presented in this work demonstrate that nanoparticulate TiO₂ in
260 exposure concentrations 10, 100 and 1000 µg TiO₂/g dry food has no effect on
261 mortality, weight change or GST activity in *P. scaber* after feeding with two sizes of
262 nanosized TiO₂ dosed food for three or 14 d. The activity of CAT and two feeding
263 parameters (food assimilation efficiency and feeding rate) were changed in dependence
264 of duration-, dose-, nanoparticle size and pretreatment (Table 4 and 5, Figs. 2-4).

265 When the animals were exposed to the same concentrations and size of
266 nanoparticles in two different experiments, the results on AE and feeding rate were not
267 entirely repeatable (at 100 µg and 1000 µg of small size TiO₂/g dry food). We explain
268 this phenomenon in the discussion.

269

270 **3.2.1 Exposure duration dependence**

271

272 After three days of exposure, there were no changes in any of measured
273 responses in animals fed on smaller nanoparticulate TiO₂ (10, 100, 1000 µg TiO₂/ g dry
274 food) when compared to the control (Table 4a, Fig. 2). However, the same exposure
275 concentrations and same type of nanoparticulate TiO₂ significantly affected CAT
276 activity and two feeding parameters after 14 d of exposure (Table 4b, Figs. 3-4).

277 When the total consumed quantities of TiO₂ were compared (experiments A and
278 B), similar total consumed quantities had different effects when ingested in 3 d or in 14
279 d. For example, a total consumed quantity in the range from 1.35 µg of TiO₂/g animal
280 wet weight (wet wt.) to 219 µg of TiO₂/g wet wt. in three days exposure had no effect

281 on measured parameters, while similar total consumed quantity (8.12 $\mu\text{g TiO}_2/\text{g wet wt.}$
282 to 905 $\mu\text{g TiO}_2/\text{g wet wt.}$) in 14 d provoked changes to some of measured parameters
283 (Table 5). These results show that the effect was not primarily related to exposure
284 concentration or total consumed quantity, but was dependent upon duration of exposure.

285

286 *3.2.2 Exposure- dose dependence*

287

288 The dose-dependent pattern for feeding parameters and CAT activity was
289 recognised to be threshold-like when animals were fed on small nanosized TiO_2 (Table
290 4b, Table 5, Fig.3). When animals were exposed to larger nano- TiO_2 no dose response
291 relationship pattern could be recognised for feeding parameters (Table 4c, Table 5, Fig.
292 4).

293

294 *3.2.3 Size of nanoparticles dependence*

295

296 When the biological effects of both sizes of nanoparticles were compared within
297 experiment C, significant differences were observed (Table 4 c, Fig. 4). Smaller
298 nanoparticles (100 $\mu\text{g/g}$ dry food exposure concentration) caused induction of feeding
299 parameters and increased CAT activity, while no change was observed at the same
300 concentration of larger nanoparticles.

301

302 *3.2.4 Pre-treatment of nanoparticles dependence*

303

304 In a simple test, we compared the effects of the same exposure concentration,
305 dose and size of nanoparticles prepared in different ways, i.e. sonicated or non-
306 sonicated (Table 4 d, Fig.5). Sonicated smaller nanoparticles of TiO₂ enhanced AE,
307 which was unaffected by the same exposure concentration of non-sonicated
308 nanoparticles. CAT activity was increased in both groups of exposed animals
309 independently on pre-treatment of nanoparticles. It is evident that the modification of
310 nanoparticles might affect their biological reactivity potential, however to what extend
311 remains to be further investigated.

312

313 **4 DISCUSSION**

314 The effects of nanosized TiO₂ on terrestrial isopods depended on the total
315 consumed quantity and exposure concentration of nanoparticles, exposure duration, and
316 the size of particles as well as their pre-treatment.

317 It was expected that the intensity of a response would reflect the amount of
318 consumed quantity of nanoparticles, but the results show that it was ruled by duration of
319 exposure and not by consumed quantity or exposure concentration of nano-TiO₂. For
320 instance, a total ingested amount of 8.12 µg/g wet wt. of smaller TiO₂ in 14 d led to
321 elevation in feeding parameters (Table 5), but the feeding parameters were not affected
322 when an even higher amount (20.4 µg/g wet wt.) of the same size TiO₂ was consumed
323 in three days.

324 As determined in the present study, the dose-response relationships for
325 nanoparticles are different from those of conventional chemicals (Drobne et al., 2008;
326 Stanek et al., 2006). Nanosized TiO₂ provoked a threshold-like dose-response of
327 parameters studied in *P. scaber*. This was evident in the case of exposure to small

328 nanosized TiO₂. Here, two orders of magnitude different concentrations of nano-TiO₂
329 had similar effect on feeding parameters and CAT activity.

330 Contrary to expectations, nanosized TiO₂ enhanced feeding rate of *P. scaber*.
331 On the basis of our previous work, we expected reduced feeding rate as recorded many
332 times upon exposure to metal dosed food (Drobne and Hopkin, 1995). We explain the
333 increase of feeding parameters as a hormetic-like response (Calabrese, 2003), which can
334 have complex time response dynamics. In the present work, this means that after 14 d of
335 exposure we can either detect an increase of the feeding response (experiment B) or
336 miss it due to its cessation (experiment D).

337 It has been suggested, that small sized particles, whose surface area per unit
338 mass is larger than that of larger particles, are more biologically potent (Borm et al.,
339 2006; Oberdörster et al., 2007; Warheit et al., 2007 b). Our results confirm this
340 suggestion. When effects of similar doses of smaller nanosized TiO₂ and larger
341 nanosized TiO₂ are compared, differences in feeding parameters and CAT activity were
342 observed (Table 4c). This might suggest different modes of action and/or
343 toxicodynamics of the two sizes of TiO₂. However, the effect of nanoparticle size
344 remains to be further studied, since the two tested sizes of nanoparticles were of
345 different crystalline phase. Smaller nano-TiO₂ particles were in pure anatase crystalline
346 phase while larger nanoparticles were a mixture of both, anatase and rutile crystalline
347 phase.

348 The effects of nanoparticles are often linked to their physico-chemical
349 characteristics (Borm et al., 2006; Oberdörster et al., 2007; Warheit et al., 2007 b). This
350 was also proven by our results. Sonicated dispersions, which formed smaller aggregates
351 than non-sonicated suspension, resulted in a higher biological potency. Anyway, the

352 measured responses observed in this work cannot be explained straightforward by size
353 and surface area of nanoparticles as analysed in aqueous dispersion. Namely, the
354 aggregation pattern of TiO₂ nanoparticles can be further changed on leaf surface and
355 inside the animal's digestive fluids due to different pH in different parts of the digestive
356 system, the presence of surfactants and other biologically active molecules (Diegoli et
357 al., 2008).

358 No adverse effects of nano-TiO₂ on isopods, such as mortality, weight change or
359 decrease of feeding, were observed in this study. Therefore the tested concentrations
360 may be considered safe for isopods exposed for three or 14 d to nanosized TiO₂.
361 Furthermore, the concentrations tested in the present study (the lowest concentration
362 was 10 µg/g dry food) are much higher as the recently reported predicted high emission
363 scenario environmental concentrations of nano-TiO₂ in soil (0.0048 µg/g) (Mueller and
364 Nowack, 2008). Other similar studies also report the low toxicity potential of nanosized
365 TiO₂ when compared to dissolved chemicals. Similar studies report the effects of TiO₂
366 on the mobility of water fleas *Daphnia magna* (no effect up to 500 mg/L) (Lovern and
367 Klaper, 2006; Warheit et al., 2007a), the mortality of crustacea *Thamnocephalus*
368 *platyurus* (no effect up to 2 g/L) (Hainlaan et al, 2008), the luminescence of bacteria
369 *Vibrio fischeri* (no effect up to 2 g/L) (Hainlaan et al., 2008), the growth of algae
370 *Pseudokirchneriella subcapitata* (72 h median effective concentration EC₅₀ = 87 mg/L)
371 (Warheit et al., 2007a), the growth of algae *Desmodesmus subspicatus* (72 h EC₅₀ = 32-
372 44 mg/L) (Hund-Rinke and Simon, 2006), and the mobility of rainbow trout
373 *Oncorhynchus mykiss* (no effect up to 100 mg/L) (Warheit et al., 2007a). It remains to
374 be further checked whether longer exposure periods, which are more realistic in the
375 field, would result in more pronounced effect of nano-TiO₂ on terrestrial isopods.

376 Despite of these data, conclusions concerning the safety of nanoparticles must be
377 drawn with great care. Safety data for nanoparticles should be interpreted as a function
378 of dose, exposure period and also size and surface modifications. To collect all these
379 data a lot of testing is needed under varying conditions and with a reasonable set of
380 endpoints. A bioassay with the terrestrial isopod *Porcellio scaber* proved to be suitable
381 for detecting effects of nanoparticles. The suite of analyzed biomarkers enables
382 detection of both early non-toxic effects as well as potential adverse effects within
383 changeable duration of exposure. Tests with isopods fit well into a set of tests suited for
384 hazard characterisation of nanoparticles (Warheit et al., 2007a).

385 **In conclusion**, the response of a biological system to nanoparticles appeared to
386 be unique and depends on the physico-chemical characteristics of nanoparticles, dose
387 and duration of exposure. The data from biological tests should therefore be interpreted
388 and processed differently from data for chemicals. This is in line with the
389 recommendations provided by the European Commission scientific committee on
390 emerging and newly identified health risks (SCENIHR, 2007). At the present state of
391 knowledge comparative information on the biological activity of nanoparticles would
392 serve best for characterization of hazard and prioritization of nanosized material.

393

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395

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399

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491 **Figure legends:**

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493 **Fig. 1:** Transmission electron micrographs of nanosized titanium dioxide (TiO₂) <25nm
494 (a) and <75nm (b) in bidestilled water (non-sonicated).

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496 **Fig. 2:** The food assimilation efficiency (a), feeding rate (b), catalase (CAT) activity (c),
497 and glutathione S-transferase activity (GST) (d) in isopods fed with small sized TiO₂
498 (<25 nm) for 3 days (Experiment A). Symbols on the box plot represent maximum and
499 minimum value (whiskers: ⊥) and mean value (■).

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501 **Fig. 3:** The food assimilation efficiency (a), feeding rate (b), catalase (CAT) activity (c),
502 and glutathione S-transferase activity (GST) (d) in isopods fed with small sized TiO₂
503 (<25 nm) for 14 days (Experiment B). Symbols on the box plot represent maximum and
504 minimum value (whiskers: ⊥) and mean value (■). The effects at a certain exposure
505 concentration, which are significantly different in comparison to control, are shown
506 (symbols denote: (*) p<0.05, and (**) p<0.001).

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508 **Fig. 4:** The food assimilation efficiency (a), feeding rate (b), catalase (CAT) activity (c),
509 and glutathione S-transferase activity (GST) (d) in isopods fed with large sized TiO₂
510 (<75 nm) for 14 days (Experiment C). Symbols on the box plot represent maximum and
511 minimum value (whiskers: ⊥) and mean value (■). The effects at a certain exposure
512 concentration, which are significantly different in comparison to control, are shown
513 (symbols denote: (±) p<0.1; and (*) p<0.05).

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516 **Fig. 5:** The food assimilation efficiency (a), feeding rate (b), catalase (CAT) activity (c),
517 and glutathione S-transferase activity (GST) (d) in isopods fed with non-sonicated and
518 sonicated small sized TiO₂ (<25 nm) for 14 days (Experiment D). Symbols on the box
519 plot represent maximum and minimum value (whiskers: ⊥) and mean value (■). The
520 effects at a certain exposure concentration, which are significantly different in
521 comparison to control, are shown (*; p<0.05).

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540 **Table 1.** Characteristics of TiO₂ nanoparticles studied in the present work

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	Small nanosized TiO₂	Large nanosized TiO₂
Supplier info (Sigma-Aldrich)	nanopowder anatase crystalline structure, particle size <25 nm, surface area 200-220 m ² /g.	amorphous liquid medium dispersion 5 wt.% in H ₂ O mixture of rutile and anatase crystalline structure, particle size <50 nm (XRD), <75 nm (BET) no data on surface area
BET (supplied material)		
particle size	10 nm	40 nm
Specific surface area	145 m ² /g	40 m ² /g
TEM (aqueous dispersion)		
Single particle size within the aggregates	10-20 nm (Fig. 1a)	10-120 nm (Fig. 1b)
Single particle shape	elongated and round	round
Description of aggregates	N - dense aggregates S – net like, loose aggregate	N - loose aggregates
DLS (aqueous dispersion)		

Size of aggregates	N – 750 to 950 nm	N – 100 to 200 nm
	S – 400 to 460 nm	

542 **Symbols:** XRD- X-ray diffraction, BET- Brunauer- Emmett -Teller surface area
543 analysis, TEM- Transmission electron micrograph, DLS- Dynamic light scatter, N: non-
544 sonicated dispersion, S-sonicated dispersion

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563 **Table 2:** Summary of the test organism, nanoparticles tested, type of exposure and
 564 endpoints evaluated in the present paper.
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Description	Endpoints evaluated	
	<i>Lower level endpoints:</i>	<i>Higher level endpoints:</i>
Test organism	Digestive glands:	- Feeding rate
Invertebrate	-glutathione S-transferase	- Food assimilation efficiency
Isopoda, Crustacea	activity	- Animal mass change
Terrestrial isopod <i>Porcellio scaber</i>	-catalase activity	- Mortality
Type of exposure		
3 d and 14 d		
dietary exposure		
Chemical		
Nano-sized TiO ₂		
< 25 nm; < 75 nm		

566 ^ad-days

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574 **Table 3:** The total number of animals exposed in each experiment

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Suspension of TiO ₂	Final exposure concentrations of TiO ₂ (µg/g dry food)	EXPERIMENTS			
		Total No. of exposed animals			
		A ^b	B	C	D
		3 d ^a	14 d ^a	14 d ^a	14 d ^a
	0	8+8+8	15	7	10
< 25 nm	10	6	15		
non-sonicated	100	8+8+7	15	9	
	1000	10+10+10+6	15		10
< 25 nm sonicated	1000				10
< 75 nm	10			7	
non-sonicated	100			9	
	1000			10	

576 ^ad-days, ^b experiment A was repeated up to 4 times, each number indicates the number
 577 of animals in each exposure

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586 **Table 4:** The effects of nanosized TiO₂ on *P. scaber*. The effects at a certain exposure
 587 concentration, which are significantly different in comparison to control, are shown.
 588 Symbols denote: (/) p>0.1-no effect, (±) p<0.1; (*) p<0.05, and (**) p<0.001).

Exp. A: 3 d			
Exposure concentration (µg/g dry food)	10	100	1000
Particle size (nm)	<25	<25	<25
AE	/	/	/
feeding rate	/	/	/
CAT	/	/	/
GST	/	/	/
weight change	/	/	/
mortality	/	/	/
Exp. B: 14 d			
Exposure concentration (µg/g dry food)	10	100	1000
Particle size (nm)	<25	<25	<25
AE	**	*	*
feeding rate	**	**	*
CAT	/	*	*
GST	/	/	/
weight change	/	/	/
mortality	/	/	/

Exp. C: 14 d				
Exposure concentration ($\mu\text{g/g}$ dry food)	10	100	1000	100
Particle size (nm)	<75	<75	<75	<25
AE	*	/	\pm	*
feeding rate	/	/	*	\pm
CAT	/	/	/	*
GST	/	/	/	/
weight change	/	/	/	
mortality	/	/	/	/

Exp. D: 14 d				
Exposure concentration ($\mu\text{g/g}$ dry food)	1000	1000		
Particle size (nm)	<25 N ^b	<25 S ^c		
AE	/	*		
feeding rate	/	/		
CAT	*	*		
GST	/	/		
weight change	/	/	/	
mortality	/	/	/	/

589 Symbols: ^bN: non-sonicated dispersion; ^cS:sonicated dispersion, d-days; AE- food

590 assimilation efficiency; CAT-catalase; GST-glutathione S-transferase

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593 **Table 5:** Comparison between the effects caused by the exposure concentrations, daily
 594 consumed doses and total consumed quantities of TiO₂. The effects at a certain exposure
 595 concentration/dose, which are significantly different in comparison to control, are
 596 shown. Symbols denote: (/) p>0.1-no effect, (±) p<0.1; (*) p<0.05, and (**) p<0.001).

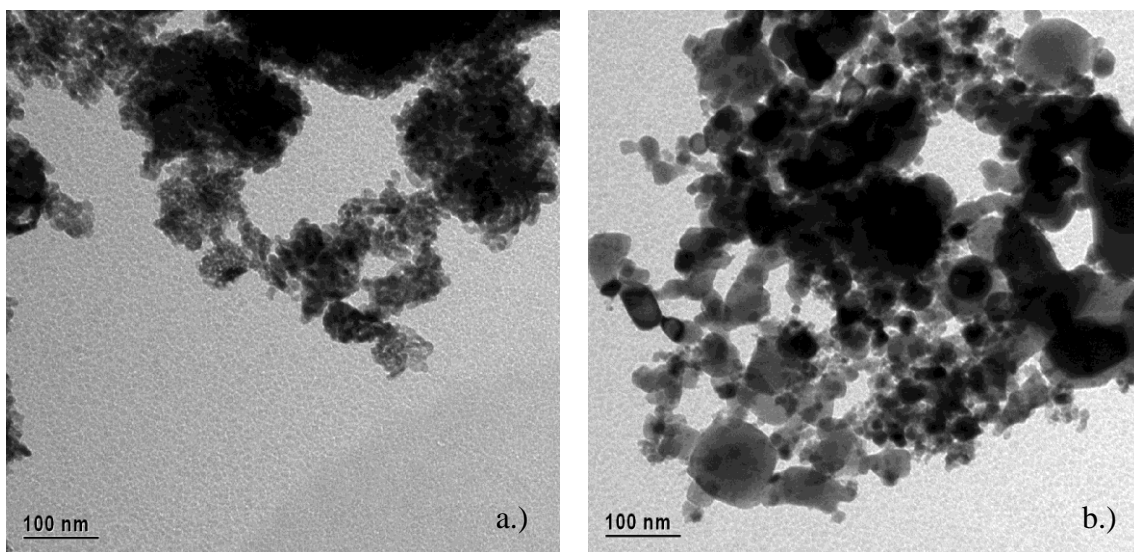
Exposure concentration of TiO ₂ (µg/g dry food)	Ex	Daily consumed dose of TiO ₂ (µg/g wet wt./day) ^a	Total consumed quantity of TiO ₂ (µg/g wet wt.) ^a	AE	Feeding	CAT
<25 nm 10 3 d	A	0.45	1.35	/	/	/
<25 nm 10 14 d	B	0.58	8.12	**	**	/
<75 nm 10 14 d	C	0.53	7.42	/	/	/
<25 nm 100 3 d	A	6.8	20.4	/	/	/
<25 nm 100 14 d	B	5.82	81.5	*	**	*
<25 nm 100 14 d	C	7.05	105	±	±	*
<75 nm 100 14 d	C	4.38	61.3	/	/	/
<25 nm 1000 3 d	A	73	219	/	/	/
<25 nm 1000 14 d	B	64.6	905.5	*	*	*
<25 nm 1000 S 14 d	D	55.7	835.5	/	/	*
<25 nm 1000 N 14 d	D	61.1	916.5	*	/	*
<75 nm 1000 14 d	C	73.21	1025	±	*	/

597 Symbols: ^a expressed per animal wet weight; ^bN: non-sonicated dispersion; ^cS:
598 sonicated, dispersion; Ex.- experiment, d-days; AE- food assimilation efficiency; CAT-
599 catalase
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601 **Figures**

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603 **Fig. 1.**



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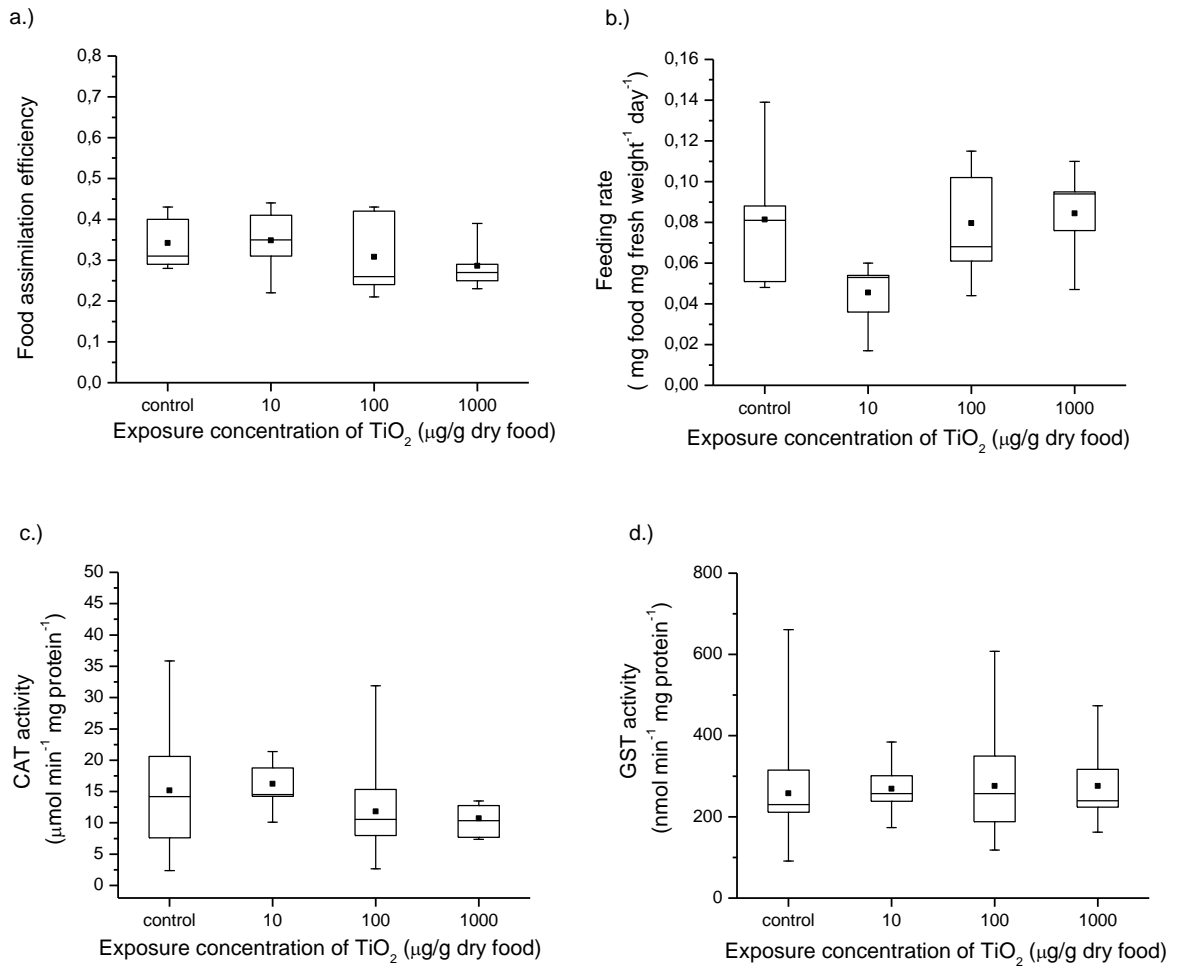
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615 **Fig. 2:**



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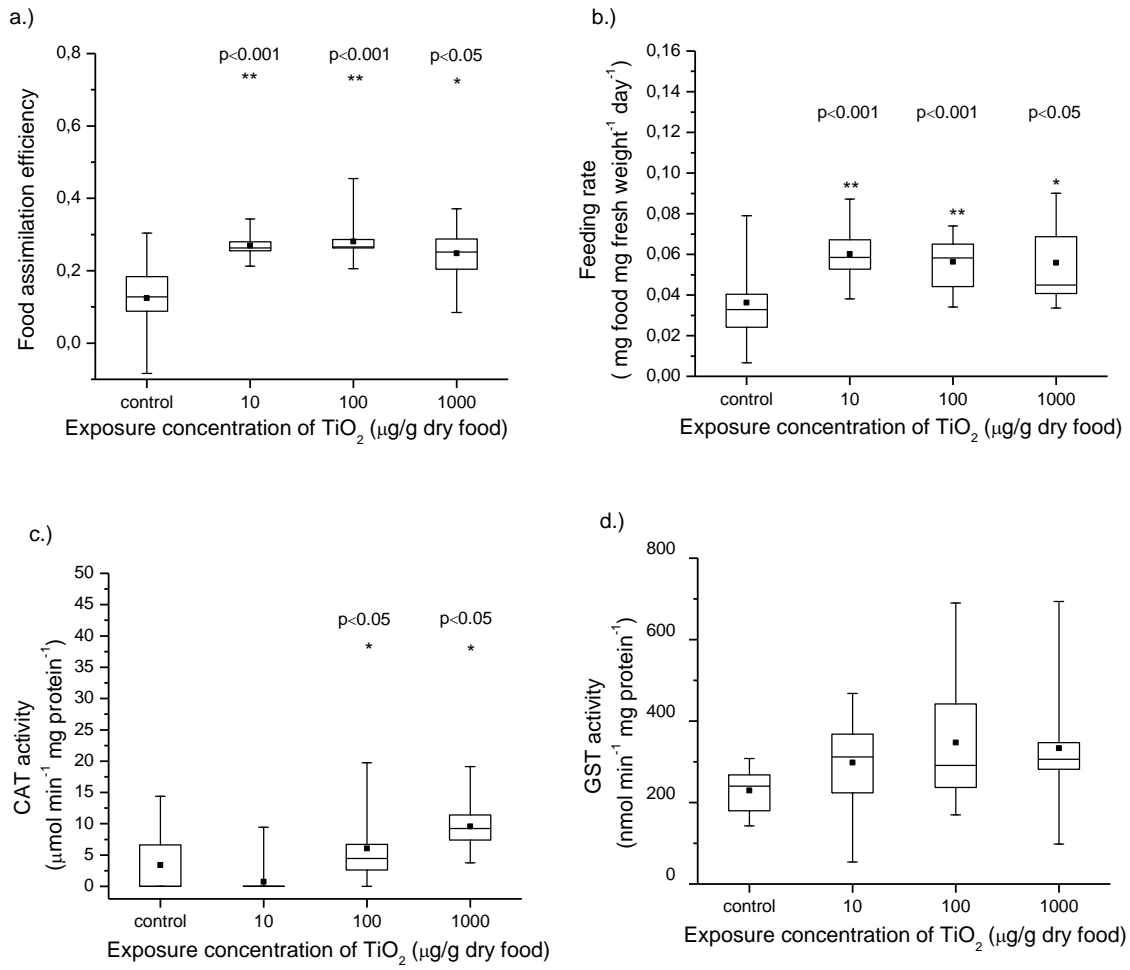
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625 **Fig.3.**



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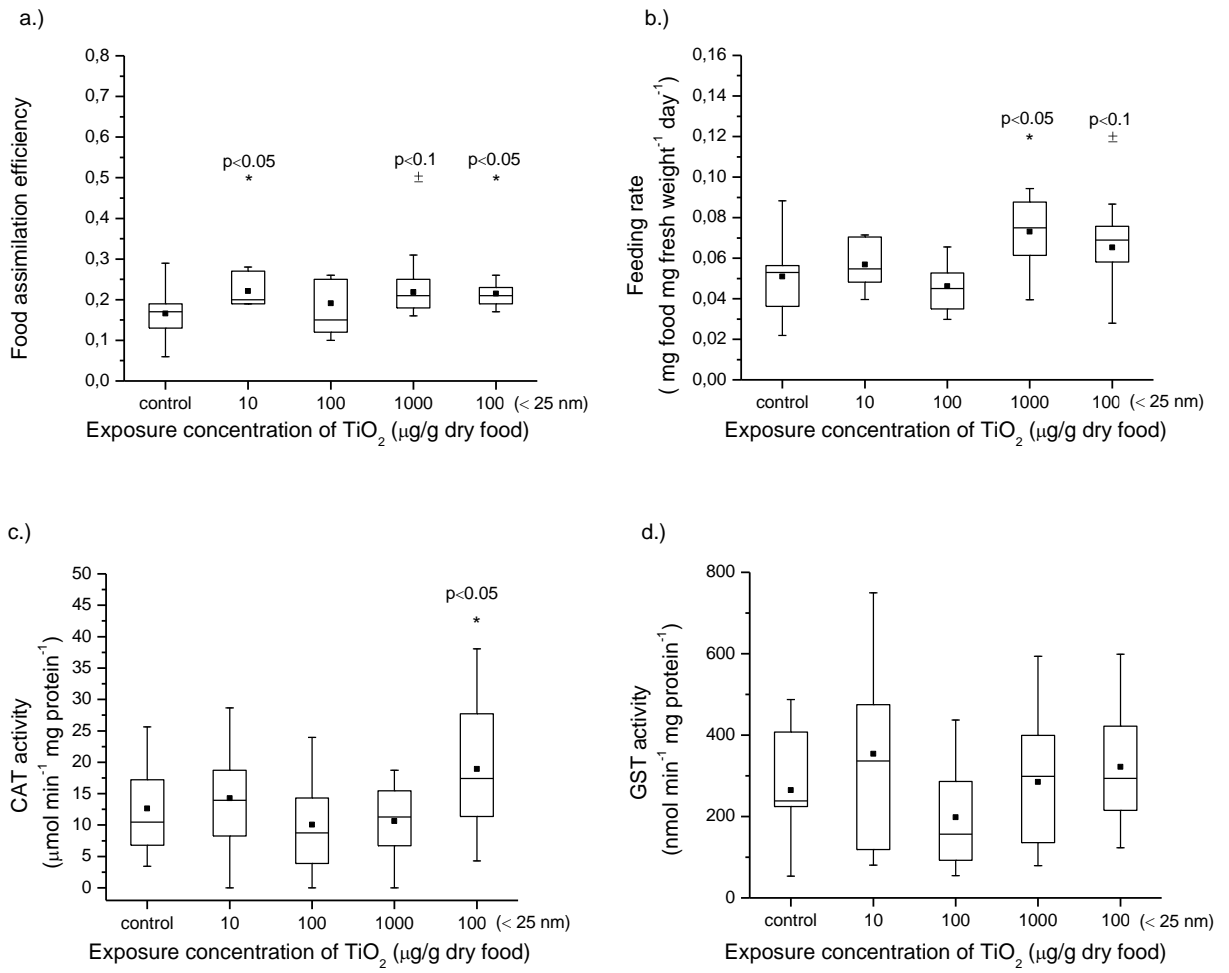
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636 **Fig. 4**



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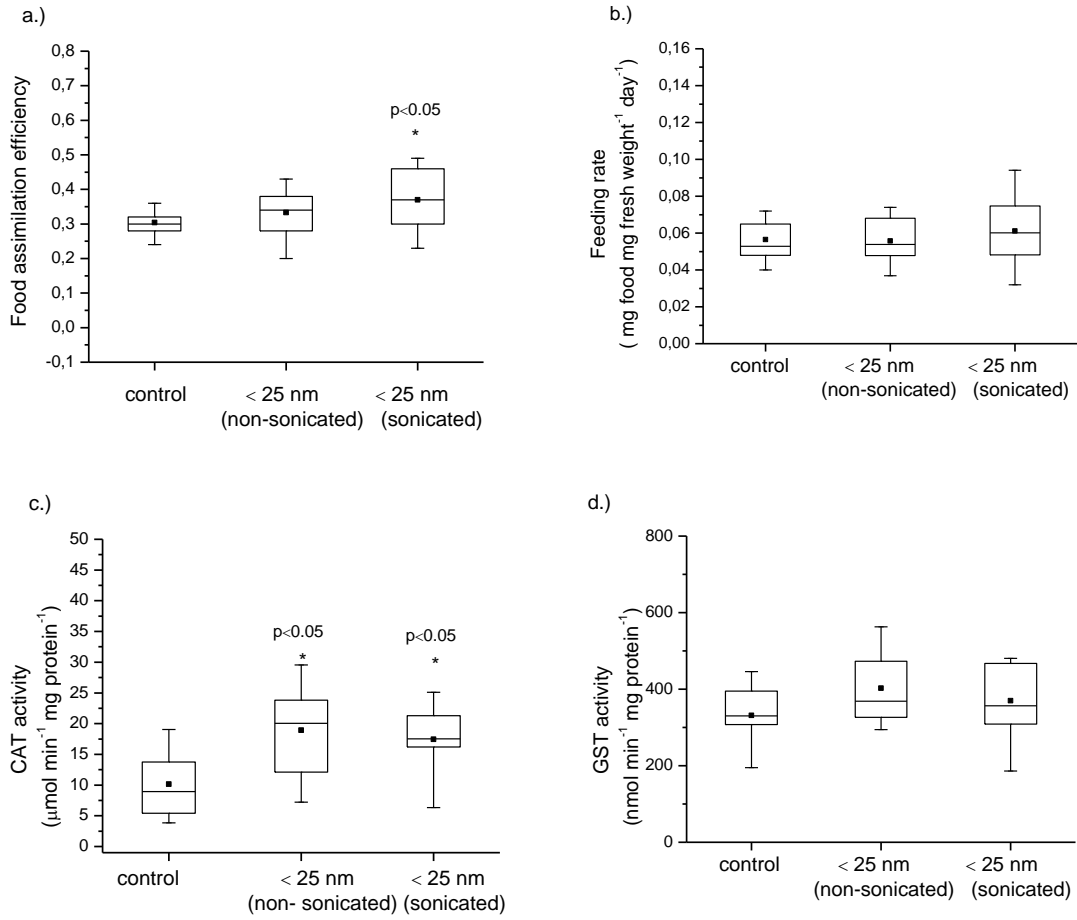
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646 **Fig. 5**

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